

L Number	Hits	Search Text	DB	Time stamp
1	792617	hydrogen sulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 17:53
2	7098	hydrogen near sulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 17:53
3	1758	hydrogensulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 17:53
4	4176	methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 17:54
5	7853	((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 18:07
6	8528	hydrogensulfate or (hydrogen near sulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 17:56
7	10945	((methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate) or ((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 17:56
8	344	((hydrogensulfate or (hydrogen near sulfate)) same ((methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate) or ((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate))	USPAT; US-PGPUB; DERWENT	2004/06/16 17:56
9	124	((hydrogensulfate or (hydrogen near sulfate)) same ((methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate) or ((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate))) and pharmaceutical	USPAT; US-PGPUB; DERWENT	2004/06/16 17:56
10	121	((hydrogensulfate or (hydrogen near sulfate)) same ((methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate) or ((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate))) and pharmaceutical) and (salt or salts)	USPAT; US-PGPUB; DERWENT	2004/06/16 17:57
29	7853	((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near (sulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 18:08
30	110	((hydrogensulfate or (hydrogen near sulfate)) same (methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 18:08
31	56	((hydrogensulfate or (hydrogen near sulfate)) same (methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate)) and pharmaceutical	USPAT; US-PGPUB; DERWENT	2004/06/16 18:09
32	56	((hydrogensulfate or (hydrogen near sulfate)) same (methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate)) and pharmaceutical) and salt	USPAT; US-PGPUB; DERWENT	2004/06/16 18:24
41	3	isopropylsulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 18:25
42	84	isopropyl near sulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 18:25
43	0	((isopropyl near sulfate) same (hydrogensulfate or (hydrogen near sulfate)))	USPAT; US-PGPUB; DERWENT	2004/06/16 18:25
44	7	((isopropyl near sulfate) and (hydrogensulfate or (hydrogen near sulfate)))	USPAT; US-PGPUB; DERWENT	2004/06/16 18:26
45	10	isopropylsulfate or ((isopropyl near sulfate) and (hydrogensulfate or (hydrogen near sulfate)))	USPAT; US-PGPUB; DERWENT	2004/06/16 18:26

L Number	Hits	Search Text	DB	Time stamp
1	792617	hydrogen sulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 17:53
2	7098	hydrogen near sulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 17:53
3	1758	hydrogensulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 17:53
4	4176	methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 17:54
5	7853	(methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 18:07
6	8528	hydrogensulfate or (hydrogen near sulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 17:56
7	10945	(methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate) or ((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 17:56
8	344	((hydrogensulfate or (hydrogen near sulfate)) same ((methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate) or ((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate)))	USPAT; US-PGPUB; DERWENT	2004/06/16 17:56
9	124	((hydrogensulfate or (hydrogen near sulfate)) same ((methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate) or ((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate))) and pharmaceutical	USPAT; US-PGPUB; DERWENT	2004/06/16 17:56
10	121	((hydrogensulfate or (hydrogen near sulfate)) same ((methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate) or ((methyl or ethyl or isopropyl or butyl or pentyl or propyl) near sulfate))) and pharmaceutical) and (salt or salts)	USPAT; US-PGPUB; DERWENT	2004/06/16 17:57
29	7853	(methyl or ethyl or isopropyl or butyl or pentyl or propyl) near (sulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 18:08
30	110	((hydrogensulfate or (hydrogen near sulfate)) same (methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 18:08
31	56	((hydrogensulfate or (hydrogen near sulfate)) same (methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate)) and pharmaceutical	USPAT; US-PGPUB; DERWENT	2004/06/16 18:09
32	56	((hydrogensulfate or (hydrogen near sulfate)) same (methylsulfate or ethylsulfate or isopropylsulfate or butylsulfate)) and pharmaceutical) and salt	USPAT; US-PGPUB; DERWENT	2004/06/16 18:11

L Number	Hits	Search Text	DB	Time stamp
1	8528	hydrogensulfate or (hydrogen near sulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 16:46
2	8130	alkyl near sulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 16:46
3	52	(hydrogensulfate or (hydrogen near sulfate)) same (alkyl near sulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 16:56
4	231	(hydrogensulfate or (hydrogen near sulfate)) and (alkyl near sulfate) and salt	USPAT; US-PGPUB; DERWENT	2004/06/16 16:57
5	250	(hydrogensulfate or (hydrogen near sulfate)) and (alkyl near sulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 16:58
6	1926	alkylsulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 16:58
7	9690	(alkyl near sulfate) or alkylsulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 16:58
8	63	(hydrogensulfate or (hydrogen near sulfate)) same ((alkyl near sulfate) or alkylsulfate)	USPAT; US-PGPUB; DERWENT	2004/06/16 16:59
9	55	((hydrogensulfate or (hydrogen near sulfate)) same ((alkyl near sulfate) or alkylsulfate)) and (salt or salts)	USPAT; US-PGPUB; DERWENT	2004/06/16 16:59
10	16	((hydrogensulfate or (hydrogen near sulfate)) same ((alkyl near sulfate) or alkylsulfate)) same (salt or salts)	USPAT; US-PGPUB; DERWENT	2004/06/16 16:59

L Number	Hits	Search Text	DB	Time stamp
1	4	((("6573381") or ("4529596") or ("4847265") or ("6429210"))).PN.	USPAT	2004/06/16 14:42
2	2	((("20030114479") or ("20030225129"))).PN.	USPAT; US-PGPUB	2004/06/16 15:22
3	597	clopidogrel	USPAT; US-PGPUB; DERWENT	2004/06/16 15:23
4	158893	l3and sulfuric	USPAT; US-PGPUB; DERWENT	2004/06/16 15:24
5	212	clopidogrel and sulfuric	USPAT; US-PGPUB; DERWENT	2004/06/16 15:24
6	678	alkyl near sulfuric	USPAT; US-PGPUB; DERWENT	2004/06/16 15:25
7	1	clopidogrel and (alkyl near sulfuric)	USPAT; US-PGPUB; DERWENT	2004/06/16 15:31
8	8130	sulfate near alkyl	USPAT; US-PGPUB; DERWENT	2004/06/16 15:31
9	8	(sulfate near alkyl) and clopidogrel	USPAT; US-PGPUB; DERWENT	2004/06/16 15:34
10	359	thienopyridine	USPAT; US-PGPUB; DERWENT	2004/06/16 15:34
11	517	thieno same pyridine	USPAT; US-PGPUB; DERWENT	2004/06/16 15:35
12	804	thienopyridine or (thieno same pyridine)	USPAT; US-PGPUB; DERWENT	2004/06/16 15:35
13	7	(sulfate near alkyl) and (thienopyridine or (thieno same pyridine))	USPAT; US-PGPUB; DERWENT	2004/06/16 15:35
14	7	((sulfate near alkyl) and (thienopyridine or (thieno same pyridine))) not ((sulfate near alkyl) and clopidogrel)	USPAT; US-PGPUB; DERWENT	2004/06/16 15:40
15	95	tetrahydrothieno same (pyridine or pyridyl)	USPAT; US-PGPUB; DERWENT	2004/06/16 15:41
16	0	(tetrahydrothieno same (pyridine or pyridyl)) and (sulfate near alkyl)	USPAT; US-PGPUB; DERWENT	2004/06/16 15:51
17	2547	hemisulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 15:51
18	1	clopidogrel same hemisulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 15:53
19	1	clopidogrel near hemisulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 15:53
20	5	clopidogrel and hemisulfate	USPAT; US-PGPUB; DERWENT	2004/06/16 16:02
21	8221	ll6 or (sulfate near alkyl)	USPAT; US-PGPUB; DERWENT	2004/06/16 16:02
22	6700	(ll6 or (sulfate near alkyl)) and salt	USPAT; US-PGPUB; DERWENT	2004/06/16 16:03
23	2431	(ll6 or (sulfate near alkyl)) same salt	USPAT; US-PGPUB; DERWENT	2004/06/16 16:03

24	447	(Il6 or (sulfate near alkyl)) near salt	USPAT; US-PGPUB; DERWENT	2004/06/16 16:04
25	1347	clopidogrel or (thienopyridine or (thieno same pyridine))	USPAT; US-PGPUB; DERWENT	2004/06/16 16:04
26	1	(clopidogrel or (thienopyridine or (thieno same pyridine))) and ((Il6 or (sulfate near alkyl)) near salt)	USPAT; US-PGPUB; DERWENT	2004/06/16 16:04

10686666

=> d his

(FILE 'HOME' ENTERED AT 16:09:07 ON 16 JUN 2004)

FILE 'REGISTRY' ENTERED AT 16:09:18 ON 16 JUN 2004

L1 STRUCTURE UPLOADED

L2 0 S L1

L3 STRUCTURE UPLOADED

L4 5 S L3

L5 76 S L3 SSS FULL

FILE 'CAPLUS' ENTERED AT 16:11:49 ON 16 JUN 2004

L6 503 S L5

FILE 'REGISTRY' ENTERED AT 16:12:36 ON 16 JUN 2004

L7 0 S L1 SUB=L5 SAMPLE

L8 0 S L1 SSS FULL SUB=L5

FILE 'CAPLUS' ENTERED AT 16:14:08 ON 16 JUN 2004

L9 83 S L6 AND SULF?

L10 46 S L9 AND SULFATE

L11 46 S L9(P) SULFATE

L12 43 S L11 AND PATENT/DT

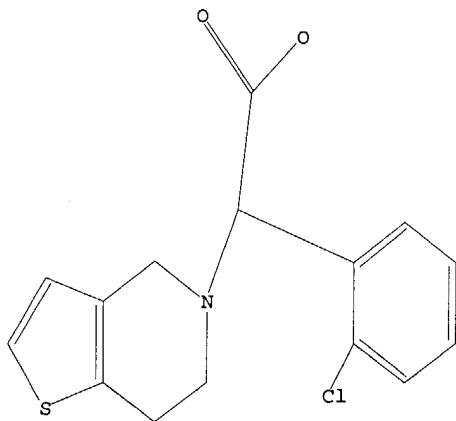
L13 22 S L12 AND HYDROCHLORIDE

L14 21 S L12 NOT L13

=> d l3

L3 HAS NO ANSWERS

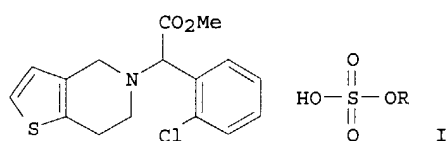
L3 STR



10686666

LI4 ANSWER 1 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2004:370683 CAPLUS
DN 140:380607
TI Preparation of clopidogrel salts with alkyl-sulphuric acids
IN Castaldi, Graziano; Bologna, Alberto; Magrone, Domenico
PA Dinamite Dipharma S.P.A. (In Abbreviated Form Dipharma S.P.A.), Italy
SO Eur. Pat. Appl., 11 pp.
CODEN: EPXXDW
DT Patent
LA English
FAN.CNT 1
PATENT NO. KIND DATE APPLICATION NO. DATE

PI EP 1415993 A1 20040506 EP 2003-23023 20031013
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
PRAI IT 2002-MI2228 A 20021021
GI



AB Clopidogrel salts with alkyl-sulfuric acids, having formula I wherein R is a straight or branched C1-C10 alkyl group; preparation thereof and the industrial and therapeutical use thereof are disclosed. A reactor was loaded at room temperature with clopidogrel hemisulfate (50 g, 0.12 mol) and isopropanol (500 mL) and refluxed under stirring. After about 5 h, the reaction mixture was cooled to room temperature and the product precipitated after approx. 3 h. The solid was filtered after about 15 h and dried under vacuum (200 mm Hg) at a temperature of 60°C for 24 h to obtain clopidogrel iso-Pr sulfate: yield = 88.8%, m.p. 167.1°C, and purity >99.9%.

IT 684269-99-6P 684270-00-6P 684270-01-7P
RL: PAC (Pharmacological activity); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)
(clopidogrel salts with alkyl-sulfuric acids)

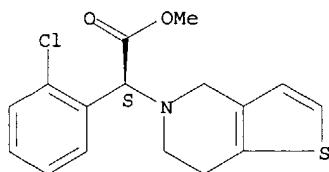
RN 684269-99-6 CAPLUS
CN INDEX NAME NOT YET ASSIGNED

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

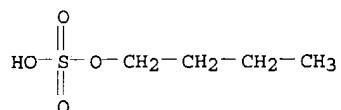
Absolute stereochemistry. Rotation (+).



CM 2

CRN 15507-13-8

CMF C4 H10 O4 S



10686666

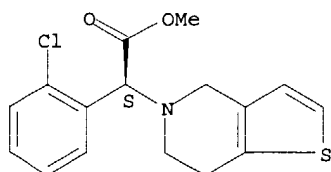
RN 684270-00-6 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, 2-methylpropyl sulfate (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

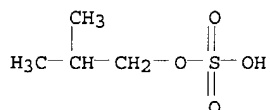
Absolute stereochemistry. Rotation (+).



CM 2

CRN 2412-30-8

CMF C4 H10 O4 S



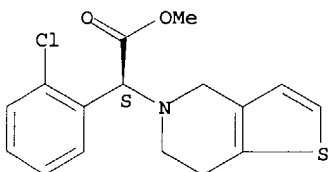
RN 684270-01-7 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, 1,1-dimethylethyl sulfate (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

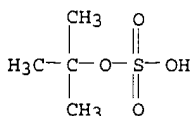
Absolute stereochemistry. Rotation (+).



CM 2

CRN 17011-26-6

CMF C4 H10 O4 S



IT 113665-84-2, Clopidogrel

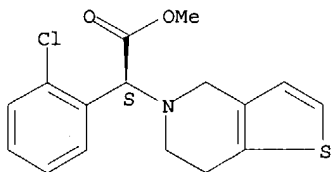
10686666

RL: RCT (Reactant); RACT (Reactant or reagent)
(clopidogrel salts with alkyl-sulfuric acids)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



IT 120202-66-6P, ClopiDogrel hemisulfate

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
(Reactant or reagent)
(clopidogrel salts with alkyl-sulfuric acids)

RN 120202-66-6 CAPLUS

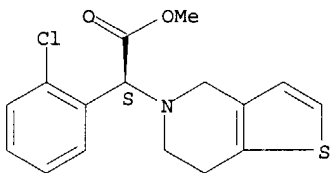
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

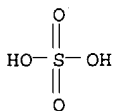
Absolute stereochemistry. Rotation (+).



CM 2

CRN 7664-93-9

CMF H2 O4 S



IT 684269-94-1P 684269-95-2P 684269-96-3P

684269-97-4P 684269-98-5P

RL: SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)
(clopidogrel salts with alkyl-sulfuric acids)

RN 684269-94-1 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, 1-methylethyl sulfate (9CI) (CA INDEX NAME)

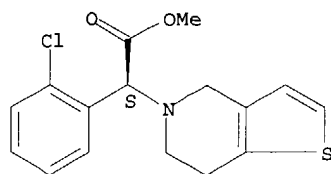
CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

Absolute stereochemistry. Rotation (+).

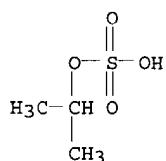
10686666



CM 2

CRN 6914-90-5

CMF C3 H8 O4 S



RN 684269-95-2 CAPLUS

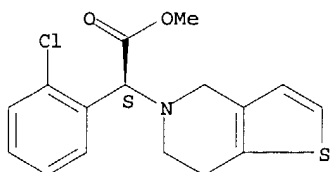
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, 1-methylpropyl sulfate (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

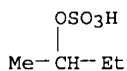
Absolute stereochemistry. Rotation (+).



CM 2

CRN 3004-76-0

CMF C4 H10 O4 S



RN 684269-96-3 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, methyl sulfate (9CI) (CA INDEX NAME)

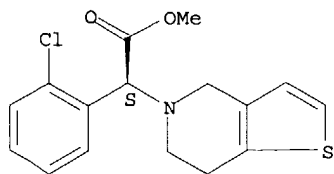
CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

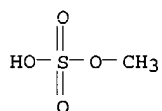
Absolute stereochemistry. Rotation (+).

10686666



CM 2

CRN 75-93-4
CMF C H4 O4 S

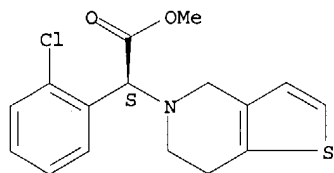


RN 684269-97-4 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, ethyl sulfate (9CI) (CA INDEX NAME)

CM 1

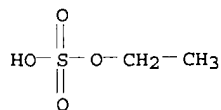
CRN 113665-84-2
CMF C16 H16 Cl N O2 S

Absolute stereochemistry. Rotation (+).



CM 2

CRN 540-82-9
CMF C2 H6 O4 S



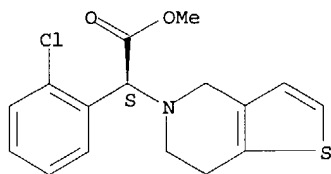
RN 684269-98-5 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, propyl sulfate (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2
CMF C16 H16 Cl N O2 S

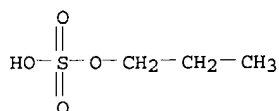
Absolute stereochemistry. Rotation (+).

10686666



CM 2

CRN 13425-84-8
CMF C3 H8 O4 S



RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

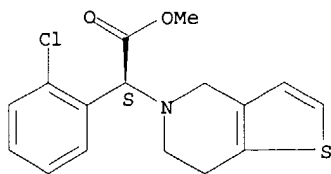
L14 ANSWER 2 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2004:333612 CAPLUS
DN 140:362998
TI Gamma irradiation of solid nanoparticulate active agents
IN Lee, Robert; Hilborn, Matthew; Kline, Laura; Keller, Janine
PA Elan Pharma International Limited, Ire.
SO PCT Int. Appl., 79 pp.
CODEN: PIXXD2
DT Patent
LA English
FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2004032980	A1	20040422	WO 2003-US27484	20030904
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU				
RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
US 2004105778	A1	20040603	US 2003-654600	20030904

PRAI US 2002-415749P P 20021004
AB The present invention relates to methods for terminal sterilization of solid forms of nanoparticulate active agent compns. via gamma irradiation The nanoparticulate active agent has an effective average particle size of less than about 2 μ , prior to incorporation into a solid form for sterilization. The resultant sterilized compns. exhibit excellent redispersibility, homogeneity, and uniformity. Also encompassed are compns. made via the described method and methods of treating animals and humans using such compns. Several examples are provided of γ -ray sterilization of naproxen nanoparticulate formulations. Pre-lyophilization, post-lyophilization and post- γ -irradiation properties (particle size, stability, osmolality, pH, microbiol. testing) are described. Surface stabilizers are used.
IT 113665-84-2, Clopidogrel
RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (γ -ray sterilization of pharmaceutical nanoparticles)
RN 113665-84-2 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).

10686666



RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 3 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2004:251975 CAPLUS
DN 140:276135
TI Method to treat collagenous connective tissue for implant remodeled by
host cells into living tissue
IN Cheung, David T.
PA USA
SO U.S. Pat. Appl. Publ., 14 pp.
CODEN: USXXCO
DT Patent
LA English
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2004057936	A1	20040325	US 2002-253017	20020923
PRAI	US 2002-253017		20020923		

AB The invention relates to a method of treatment of collagenous connective tissue removed from a donor for implant into a recipient which is re-habited or re-colonized by host cells without an immune rejection and inflammatory reaction. After removal from the donor, the tissue is trimmed and thereafter soaked in a cold stabilizing solution having a temperature range of 4 to 10 °C. The tissue is then soaked at a predetd. temperature in a polyglycol, salt, hydrogen peroxide, and phosphate buffer first solution of predetd. quantities and concns. and of sufficient ionic strength to permit ground substances to dissociate such that the collagen fibers remain stable. The tissue is then soaked in an alc. and water solution at a predetd. temperature for a sufficient period of time to remove the residue of the first solution. Following the removal of the residue, the tissue is soaked at a predetd. temperature in a third solution of an anti-inflammatory agent, an anti-thrombic agent, alc., and water or sequentially in an anti-inflammatory agent, alc., and water solution, and then in an anti-thrombic agent, alc. and water solution and thereafter stored.

IT 113665-84-2, Clopidogrel

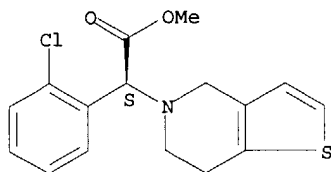
RL: BUU (Biological use, unclassified); BIOL (Biological study); USES (Uses)

(method to treat collagenous connective tissue for implants remodeled by host cells into living tissue)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



L14 ANSWER 4 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2004:203837 CAPLUS
DN 140:241063
TI Method for the manufacture of crystalline form I of clopidogrel hydrogen
sulfate
IN Veverka, Miroslav; Vodny, Stefan; Veverkova, Eva; Hajicek, Josef;
Stepankova, Hana
PA Leciva, A.S., Czech Rep.
SO PCT Int. Appl., 18 pp.

10686666

CODEN: PIXXD2

DT Patent
LA English
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2004020443	A1	20040311	WO 2003-CZ49	20030826
	W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU			
	RW:	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG			

PRAI CZ 2002-2906 A 20020827

AB A method for manufacturing the hydrogen sulfate (alpha S) of the alpha-(2-chlorophenyl)-6,7-dihydrothieno[3,2-c]pyridine-5(4H)-acetic acid Me ester (i.e., clopidogrel hydrogen sulfate), in crystalline Form I, where the compound is separated out of a solution of clopidogrel in the form of the free base or salt in a solvent selected from the series of primary, secondary or tertiary C1-5 alcs. (e.g., 2-propanol), their esters with C1-4 carboxylic acids, or optionally of mixts. thereof.

IT 120202-66-6P, Clopidogrel hydrogen sulfate
RL: PEP (Physical, engineering or chemical process); PRP (Properties); PYP (Physical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)
(method for the manufacture of crystalline form I of clopidogrel hydrogen sulfate)

RN 120202-66-6 CAPLUS

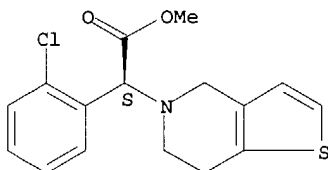
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, alpha-(2-chlorophenyl)-6,7-dihydro-, methyl ester, (alphaS)-, sulfate (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

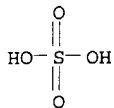
Absolute stereochemistry. Rotation (+).



CM 2

CRN 7664-93-9

CMF H2 O4 S



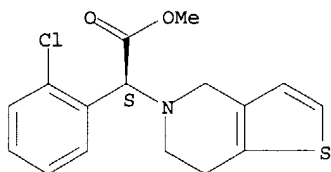
IT 113665-84-2, Clopidogrel
RL: RCT (Reactant); RACT (Reactant or reagent)
(method for the manufacture of crystalline form I of clopidogrel hydrogen sulfate using)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, alpha-(2-chlorophenyl)-6,7-dihydro-, methyl ester, (alphaS)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).

10686666



RE.CNT 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 5 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2004:60341 CAPLUS
DN 140:117406
TI Liquid dosage compositions of stable nanoparticulate drugs
IN Bosch, William H.; Hilborn, Matthew R.; Hovey, Douglas C.; Kline, Laura
J.; Lee, Robert W.; Pruitt, John D.; Ryde, Niels P.; Ryde, Tuula A.; Xu,
Shuqian
PA Elan Pharma International, Ltd, Ire.
SO PCT Int. Appl., 68 pp.
CODEN: PIXXD2
DT Patent
LA English
FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2004006959	A1	20040122	WO 2003-US22187	20030716
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU				
RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				

PRAI US 2002-396530P P 20020716

AB The present invention relates to liquid dosage compns. of stable nanoparticulate drugs. The liquid dosage compns. of the invention include osmotically active crystal growth inhibitors that stabilize the nanoparticulate active agents against crystal and particle size growth of the drug. Thus, an aqueous nanoparticulate colloidal dispersion (NCD) comprising drug 32.5 Copovidone 6.5, and dioctyl sodium sulfosuccinate 0.464% by weight was prepared by milling for 3.8 h under high energy milling conditions. The final mean particle size (by weight) of the drug particles was 161 nm. The concentrated NCD was then diluted with preserved water and glycerol (the osmotically active crystal growth inhibitor) to 0.5-3.0% drug.

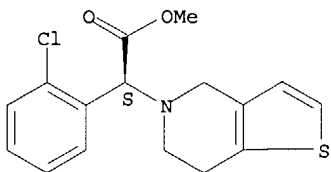
IT 113665-84-2, Clopidogrel

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
(liquid dosage compns. of stable nanoparticulate drugs)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 6 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

10686666

AN 2003:950061 CAPLUS
 DN 140:8764
 TI Polymorphs of clopidogrel hydrogen **sulfate**
 IN Lifshitz-Liron, Revital; Kovalevski-Ishai, Eti; Wizel, Shlomit;
 Avhar-Maydan, Sharon; Lidor-Hadas, Rami
 PA Israel
 SO U.S. Pat. Appl. Publ., 31 pp., Cont.-in-part of U.S. Ser. No. 74,409.
 CODEN: USXXCO
 DT **Patent**
 LA English
 FAN.CNT 3

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2003225129	A1	20031204	US 2003-339008	20030108
	US 2003114479	A1	20030619	US 2002-74409	20020212
	WO 2003051362	A2	20030626	WO 2002-US40679	20021218
	WO 2003051362	A3	20030807		
	W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, ME, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			
	RW:	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG			

PRAI US 2002-348182P P 20020111
 US 2002-74409 A2 20020212
 US 2002-359157P P 20020221
 WO 2002-US40679 A 20021218
 US 2001-342440P P 20011218
 US 2001-342351P P 20011221

AB Provided are new crystalline Forms III, IV, V and VI of clopidogrel hydrogen **sulfate** and the amorphous form of clopidogrel hydrogen **sulfate**, as well as their pharmaceutical compns., and method of treatments with such compns. Also provided are novel processes for the preparation of clopidogrel hydrogen **sulfate** Form I, Form II, Form III, Form IV, Form V, Form VI and amorphous form. Clopidogrel base (4.27 g) was dissolved in Me Et ketone (MEK) (33.7 mL). Eighty percent aqueous H2SO4 (1.03 mL) was added to the solution at 20°. The reaction mixture was heated to reflux temperature for 2 h and then the solution was cooled to room temperature and stirred at this temperature for addnl. 67 h during which a precipitate was formed. The white solid was collected by filtration, washed with MEK and dried at 50° in a vacuum oven for 24 h to obtain 4.59 g (82%) of clopidogrel hydrogen **sulfate** crystal Form II.

IT 120202-66-6P, Clopidogrel hydrogen **sulfate**
 RL: PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use);
 BIOL (Biological study); PREP (Preparation); USES (Uses)
 (polymorphs of clopidogrel hydrogen **sulfate**)

RN 120202-66-6 CAPLUS

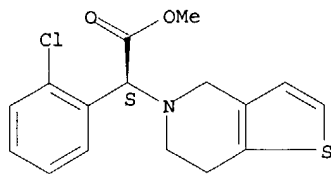
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

Absolute stereochemistry. Rotation (+).

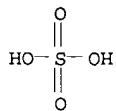


CM 2

CRN 7664-93-9

CMF H2 O4 S

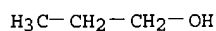
10686666



IT 548771-51-3
RL: PRP (Properties); THU (Therapeutic use); BIOL (Biological study); USES
(Uses)
(polymorphs of clopidogrel hydrogen sulfate)
RN 548771-51-3 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-
dihydro-, methyl ester, (α S)-, sulfate, compd. with 1-propanol
(1:1:?) (9CI) (CA INDEX NAME)

CM 1

CRN 71-23-8
CMF C3 H8 O



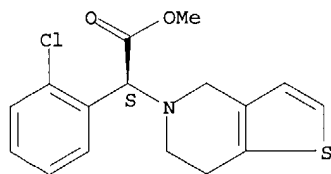
CM 2

CRN 120202-66-6
CMF C16 H16 Cl N O2 S . H2 O4 S

CM 3

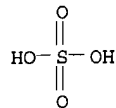
CRN 113665-84-2
CMF C16 H16 Cl N O2 S

Absolute stereochemistry. Rotation (+).



CM 4

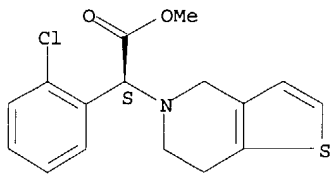
CRN 7664-93-9
CMF H2 O4 S



IT 113665-84-2, Clopidogrel
RL: RCT (Reactant); THU (Therapeutic use); BIOL (Biological study); RACT
(Reactant or reagent); USES (Uses)
(polymorphs of clopidogrel hydrogen sulfate)
RN 113665-84-2 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-
dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).

10686666



L14 ANSWER 7 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2003:491043 CAPLUS

DN 139:74015

TI Polymorphs of clopidogrel hydrogen **sulfate**

IN Lifshitz-Liron, Revital; Kovalevski-Ishai, Eti; Wizel, Shlomit;
Avhar-Maydan, Sharon; Lidor-Hadas, Rami

PA Teva Pharmaceutical Industries Ltd., Israel; Teva Pharmaceuticals USA,
Inc.

SO PCT Int. Appl., 70 pp.

CODEN: PIXXD2

DT **Patent**

LA English

FAN.CNT 3

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2003051362	A2	20030626	WO 2002-US40679	20021218
	WO 2003051362	A3	20030807		
	W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			
	RW:	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG			
	US 2003114479	A1	20030619	US 2002-74409	20020212
	US 2003225129	A1	20031204	US 2003-339008	20030108
PRAI	US 2001-342440P	P	20011218		
	US 2001-342351P	P	20011221		
	US 2002-348182P	P	20020111		
	US 2002-74409	A	20020212		
	US 2002-359157P	P	20020221		
	WO 2002-US40679	A	20021218		

AB Provided are new crystalline Forms III, IV, V and VI of clopidogrel hydrogen **sulfate** and the amorphous form of clopidogrel hydrogen **sulfate**, as well as their pharmaceutical compns. for inhibiting platelet aggregation. Also provided are novel processes for preparation of clopidogrel hydrogen **sulfate** Form I, Form II, Form III, Form IV, Form V, Form VI and amorphous form. For example, 5.31 g of clopidogrel base was dissolved in 41.9 mL of Et acetate, and 1.29 mL of 80% aqueous H₂SO₄ was added. The reaction mixture was heated and a massive precipitate was formed; the solution was cooled to room temperature, and white solid was collected by filtration, washed with Et acetate and dried to obtain 4.60 g (66%) clopidogrel hydrogen **sulfate** Form II.

IT 548771-48-8 548771-49-9 548771-50-2

548771-51-3

RL: FMU (Formation, unclassified); PRP (Properties); THU (Therapeutic use); BIOL (Biological study); FORM (Formation, nonpreparative); USES (Uses)

(preparation of amorphous and polymorphic forms of clopidogrel hydrogen **sulfate** for inhibition of platelet aggregation)

RN 548771-48-8 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate, compd. with 1-butanol (1:1:?) (9CI) (CA INDEX NAME)

CM 1

CRN 71-36-3

CMF C4 H10 O

H₃C-CH₂-CH₂-CH₂-OH

10686666

CM 2

CRN 120202-66-6

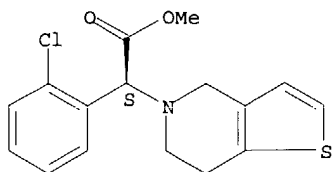
CMF C16 H16 Cl N O2 S . H2 O4 S

CM 3

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

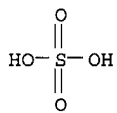
Absolute stereochemistry. Rotation (+).



CM 4

CRN 7664-93-9

CMF H2 O4 S



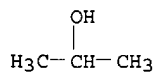
RN 548771-49-9 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate, compd. with 2-propanol (1:1:?) (9CI) (CA INDEX NAME)

CM 1

CRN 67-63-0

CMF C3 H8 O



CM 2

CRN 120202-66-6

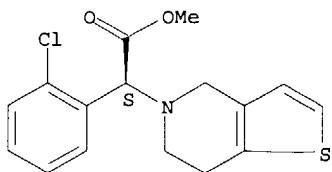
CMF C16 H16 Cl N O2 S . H2 O4 S

CM 3

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

Absolute stereochemistry. Rotation (+).

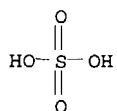


10686666

CM 4

CRN 7664-93-9

CMF H2 O4 S



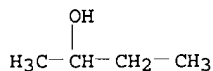
RN 548771-50-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate, compd. with 2-butanol (1:1:?) (9CI) (CA INDEX NAME)

CM 1

CRN 78-92-2

CMF C4 H10 O



CM 2

CRN 120202-66-6

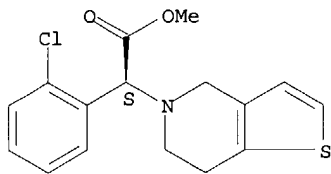
CMF C16 H16 Cl N O2 S . H2 O4 S

CM 3

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

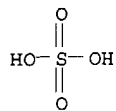
Absolute stereochemistry. Rotation (+).



CM 4

CRN 7664-93-9

CMF H2 O4 S



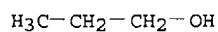
RN 548771-51-3 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate, compd. with 1-propanol (1:1:?) (9CI) (CA INDEX NAME)

CM 1

10686666

CRN 71-23-8
CMF C3 H8 O



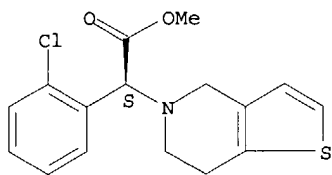
CM 2

CRN 120202-66-6
CMF C16 H16 Cl N O2 S . H2 O4 S

CM 3

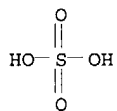
CRN 113665-84-2
CMF C16 H16 Cl N O2 S

Absolute stereochemistry. Rotation (+).



CM 4

CRN 7664-93-9
CMF H2 O4 S



IT 120202-66-6P, Clopidogrel hydrogen **sulfate**
RL: PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use);
BIOL (Biological study); PREP (Preparation); USES (Uses)
(preparation of amorphous and polymorphic forms of clopidogrel hydrogen
sulfate for inhibition of platelet aggregation)

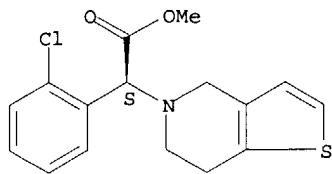
RN 120202-66-6 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-
dihydro-, methyl ester, (α S)-, sulfate (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2
CMF C16 H16 Cl N O2 S

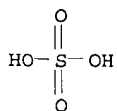
Absolute stereochemistry. Rotation (+).



CM 2

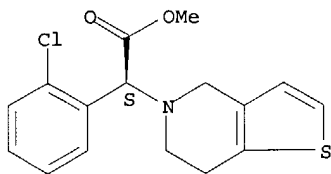
10686666

CRN 7664-93-9
CMF H2 O4 S



IT 113665-84-2, Clopidogrel
RL: RCT (Reactant); RACT (Reactant or reagent)
(preparation of amorphous and polymorphic forms of clopidogrel hydrogen
sulfate for inhibition of platelet aggregation)
RN 113665-84-2 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-
dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



L14 ANSWER 8 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2003:473265 CAPLUS
DN 139:41853
TI preparation of crystal and amorphous forms of clopidogrel hydrogen
sulfate for pharmaceuticals
IN Lifshitz-Liron, Revital; Kovalevski-Ishai, Eti; Wizel, Shlomit; Maydan,
Sharon Avhar; Lidor-Hadas, Rami
PA Israel
SO U.S. Pat. Appl. Publ., 27 pp.
CODEN: USXXCO
DT Patent
LA English
FAN.CNT 3

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2003114479	A1	20030619	US 2002-74409	20020212
	WO 2003051362	A2	20030626	WO 2002-US40679	20021218
	WO 2003051362	A3	20030807		
	W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			
	RW:	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG			
	US 2003225129	A1	20031204	US 2003-339008	20030108
PRAI	US 2001-342440P	P	20011218		
	US 2001-342351P	P	20011221		
	US 2002-348182P	P	20020111		
	US 2002-74409	A	20020212		
	US 2002-359157P	P	20020221		
	WO 2002-US40679	A	20021218		

AB The present invention provides new crystalline forms III, IV and V of clopidogrel hydrogen sulfate and the amorphous form of clopidogrel hydrogen sulfate, as well as their pharmaceutical compns., and method of treatments with such compns. The present invention further provides a novel process where the amorphous form is converted to Form I by contacting Form I with an ether. Clopidogrel hydrogen sulfate (2 g) was dissolved in MeOH (4 mL). The resulting solution was added dropwise to di-Et ether (350 mL). The suspension was stirred at

10686666

room temperature for 45 min. The solid was filtered and dried at about 50° in a vacuum oven for 24 h to give 1.12 g (56%) of clopidogrel hydrogen **sulfate**, which characterization data showed to be the amorphous form.

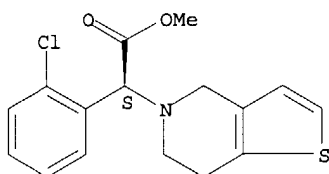
IT 120202-66-6P, Clopidogrel hydrogen **sulfate**
RL: PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use);
BIOL (Biological study); PREP (Preparation); USES (Uses)
(preparation of crystal and amorphous forms of clopidogrel hydrogen **sulfate** for pharmaceuticals)
RN 120202-66-6 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

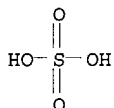
Absolute stereochemistry. Rotation (+).



CM 2

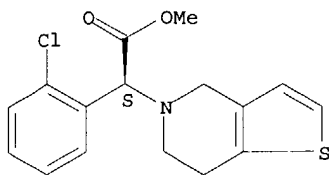
CRN 7664-93-9

CMF H2 O4 S



IT 113665-84-2, Clopidogrel
RL: RCT (Reactant); THU (Therapeutic use); BIOL (Biological study); RACT (Reactant or reagent); USES (Uses)
(preparation of crystal and amorphous forms of clopidogrel hydrogen **sulfate** for pharmaceuticals)
RN 113665-84-2 CAPLUS
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



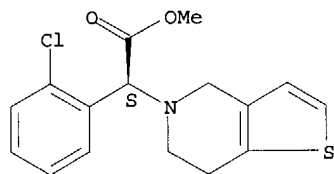
L14 ANSWER 9 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2003:334829 CAPLUS
DN 138:343889
TI Novel pharmaceutical compounds containing drugs bound to polypeptides
IN Picariello, Thomas
PA New River Pharmaceuticals Inc., USA
SO PCT Int. Appl., 4662 pp.
CODEN: PIXXD2
DT **Patent**

10686666

LA English
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2003034980	A2	20030501	WO 2001-US43089	20011114
	WO 2003034980	C1	20031120		
	W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			
	RW:	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG			
	EP 1401374	A1	20040331	EP 2001-274606	20011114
	R:	AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR			
PRAI	US 2000-274622P	P	20001114		
	WO 2001-US43089	W	20011114		
AB	Compns. comprising polypeptides and drugs covalently attached to the polypeptide are disclosed. Also provided is a method for delivery of these drugs to a patient comprising administering to the patient a composition comprising a polypeptide and a drug covalently attached to the polypeptide. Also provided is a method for protecting drugs from degradation comprising covalently attaching them to a polypeptide. Also provided is a method for controlling release of drugs from a composition comprising covalently attaching them to the polypeptide.				
IT	113665-84-2DP, Clopidogrel, protein conjugates RL: PNU (Preparation, unclassified); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses) (novel pharmaceutical compds. containing drugs bound to polypeptides)				
RN	113665-84-2 CAPLUS				
CN	Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)				

Absolute stereochemistry. Rotation (+).



L14 ANSWER 10 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2003:319495 CAPLUS
DN 138:343864
TI In vivo delivery methods and compositions
IN Kensey, Kenneth
PA USA
SO U.S. Pat. Appl. Publ., 45 pp., Cont.-in-part of U.S. Ser. No. 819,924.
CODEN: USXXCO
DT Patent
LA English
FAN.CNT 7

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2003078517	A1	20030424	US 2001-839785	20010420
	US 6019735	A	20000201	US 1997-919906	19970828
	CA 2301161	AA	19990304	CA 1998-2301161	19980826
	NZ 502905	A	20010831	NZ 1998-502905	19980826
	JP 2001514384	T2	20010911	JP 2000-507994	19980826
	US 6322524	B1	20011127	US 1999-439795	19991112
	US 6322525	B1	20011127	US 2000-501856	20000210
	NO 2000000944	A	20000225	NO 2000-944	20000225
	US 6428488	B1	20020806	US 2000-615340	20000712
	US 2001039828	A1	20011115	US 2001-789350	20010221
	US 2002007664	A1	20020124	US 2001-897164	20010702
	US 6484565	B2	20021126		
	WO 2002043806	A2	20020606	WO 2001-US44352	20011127
	WO 2002043806	A3	20030327		
	W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,			

CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
 GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
 LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL,
 PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG,
 UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
 RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY,
 DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF,
 BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
 AU 2002026986 A5 20020611 AU 2002-26986 20011127
 US 2002088953 A1 20020711 US 2001-33841 20011227
 US 6624435 B2 20030923
 WO 2002079778 A2 20021010 WO 2002-US3984 20020207
 WO 2002079778 A3 20030710

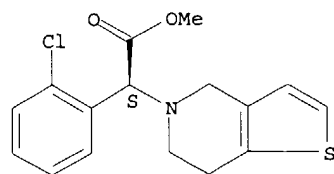
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
 CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
 GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
 LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL,
 PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG,
 UZ, VN, YU, ZA, ZW
 RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AM, AZ, BY, KG,
 KZ, MD, RU, TJ, TM, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR,
 IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN,
 GQ, GW, ML, MR, NE, SN, TD, TG

US 2002184941 A1 20021212 US 2002-156165 20020528
 US 6571608 B2 20030603
 PRAI US 1997-919906 A2 19970828
 US 1999-439795 A2 19991112
 US 2000-501856 A2 20000210
 US 2000-628401 A2 20000801
 US 2000-727950 B2 20001201
 US 2001-819924 A2 20010328
 US 1997-966076 A 19971107
 WO 1998-US17657 W 19980826
 KR 2000-16044 A 20000329
 US 2000-615340 A3 20000712
 US 2000-228612P P 20000828
 US 2001-789350 A2 20010221
 US 2001-828761 A 20010409
 US 2001-839785 A 20010420
 US 2001-841389 A 20010424
 US 2001-897164 A3 20010702
 WO 2001-US44352 W 20011127

AB Various methods are provided for determining and utilizing the viscosity of the circulating blood of a living being over a range of shear rates for diagnostics and treatment, such as detecting/reducing blood viscosity, work of the heart, contractility of the heart, for detecting/reducing the surface tension of the blood, for detecting plasma viscosity, for explaining/countering endothelial cell dysfunction, for providing high and low blood vessel wall shear stress data, red blood cell deformability data, lubricity of blood, and for treating different ailments such as peripheral arterial disease in combination with administering to a living being at least 1 drug. Agents effective to regulate at least 1 of the aforementioned blood parameters are used to adjust distribution of a substance through the bloodstream.

IT 113665-84-2, Clopidogrel
 RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
 (in vivo delivery methods and compns.)
 RN 113665-84-2 CAPLUS
 CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



10686666

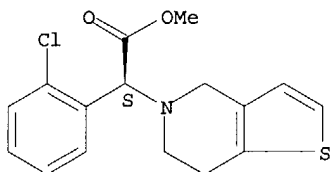
TI Combination of an ADP-receptor blocking antiplatelet drug and a
thromboxane A2 receptor antagonist for inhibiting thrombus formation
IN Ogletree, Martin L.
PA Bristol-Myers Squibb Company, USA
SO U.S., 20 pp.
CODEN: USXXAM

DT Patent
LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 6509348	B1	20030121	US 1999-428611	19991027
	US 2003109543	A1	20030612	US 2002-295347	20021115
PRAI	US 1998-106813P	P	19981103		
	US 1999-428611	A3	19991027		
AB	A method is provided for inhibiting platelet aggregation and thrombus formation by administering to a patient a synergistic combination of an ADP-receptor blocking antiplatelet drug, such as clopidogrel, with a thromboxane A2 receptor antagonist, such as ifetroban, and optionally a cholesterol lowering drug, such as an HMG CoA reductase inhibitor, for example, pravastatin. Capsules containing 98 mg clopidogrel hydrogen sulfate and 35 mg ifetroban were prepared				
IT	113665-84-2, Clopidogrel 120202-66-6, Clopidogrel hydrogen sulfate RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses) (combination of ADP-receptor blocking antiplatelet drug and thromboxane A2 receptor antagonist for inhibiting thrombus formation)				
RN	113665-84-2 CAPLUS				
CN	Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)				

Absolute stereochemistry. Rotation (+).



RN 120202-66-6 CAPLUS

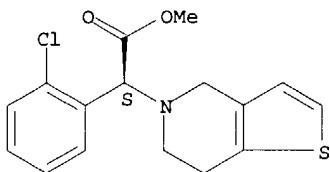
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

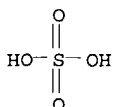
Absolute stereochemistry. Rotation (+).



CM 2

CRN 7664-93-9

CMF H2 O4 S



10686666

RE.CNT 13 THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 12 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2002:392219 CAPLUS
DN 136:406945
TI Methods for in vivo drug delivery based on monitoring blood flow
parameters
IN Kensey, Kenneth R.
PA USA
SO U.S. Pat. Appl. Publ., 40 pp., Cont.-in-part of U.S. Ser. No. 727,950.
CODEN: USXXCO
DT Patent
LA English
FAN.CNT 7

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2002061835	A1	20020523	US 2001-828761	20010409
	US 6019735	A	20000201	US 1997-919906	19970828
	CA 2301161	AA	19990304	CA 1998-2301161	19980826
	NZ 502905	A	20010831	NZ 1998-502905	19980826
	JP 2001514384	T2	20010911	JP 2000-507994	19980826
	US 6322524	B1	20011127	US 1999-439795	19991112
	US 6322525	B1	20011127	US 2000-501856	20000210
	NO 2000000944	A	20000225	NO 2000-944	20000225
	US 6428488	B1	20020806	US 2000-615340	20000712
	US 2001039828	A1	20011115	US 2001-789350	20010221
	US 2002007664	A1	20020124	US 2001-897164	20010702
	US 6484565	B2	20021126		
	WO 2002043806	A2	20020606	WO 2001-US44352	20011127
	WO 2002043806	A3	20030327		
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
	RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
	AU 2002026986	A5	20020611	AU 2002-26986	20011127
	US 2002088953	A1	20020711	US 2001-33841	20011227
	US 6624435	B2	20030923		
	WO 2002079778	A2	20021010	WO 2002-US3984	20020207
	WO 2002079778	A3	20030710		
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW				
	RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
	US 2002184941	A1	20021212	US 2002-156165	20020528
	US 6571608	B2	20030603		
PRAI	US 1997-919906	A2	19970828		
	US 1999-439795	A2	19991112		
	US 2000-501856	A2	20000210		
	US 2000-628401	A2	20000801		
	US 2000-727950	A2	20001201		
	US 1997-966076	A	19971107		
	WO 1998-US17657	W	19980826		
	KR 2000-16044	A	20000329		
	US 2000-615340	A3	20000712		
	US 2000-228612P	P	20000828		
	US 2001-789350	A2	20010221		
	US 2001-819924	A	20010328		
	US 2001-828761	A	20010409		
	US 2001-839785	A	20010420		
	US 2001-841389	A	20010424		
	US 2001-897164	A3	20010702		
	WO 2001-US44352	W	20011127		
AB	Various methods are provided for determining and utilizing the viscosity of the circulating blood of a living being over a range of shear rates for diagnostics and treatment, such as detecting/reducing blood viscosity,				

10686666

work of the heart, contractility of the heart, for detecting/reducing the surface tension of the blood, for detecting plasma viscosity, for explaining/countering endothelial cell dysfunction, for providing high and low blood vessel wall shear stress data, red blood cell deformability data, lubricity of blood, and for treating different ailments such as peripheral arterial disease in combination with administering to a living being at least one pharmaceutically acceptable agent. Agents pharmaceutically effective to regulate at least one of the aforementioned blood parameters are used to adjust distribution of a substance through the bloodstream.

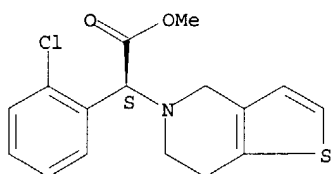
IT 113665-84-2, Clopidogrel

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
(methods for in vivo drug delivery based on monitoring blood flow parameters)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



L14 ANSWER 13 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2002:185694 CAPLUS

DN 136:252483

TI Clear oil-containing pharmaceutical compositions containing a therapeutic agent

IN Chen, Feng-Jing; Patel, Mahesh V.; Fikstad, David T.

PA USA

SO U.S. Pat. Appl. Publ., 45 pp., Cont.-in-part of U.S. Ser. No. 751,968.

CODEN: USXXCO

DT Patent

LA English

FAN.CNT 12

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2002032171	A1	20020314	US 2001-877541	20010608
	US 6267985	B1	20010731	US 1999-345615	19990630
	US 6309663	B1	20011030	US 1999-375636	19990817
	US 2001024658	A1	20010927	US 2000-751968	20001229
	US 6458383	B2	20021001		
	US 2003077297	A1	20030424	US 2002-74687	20020211
	US 2003104048	A1	20030605	US 2002-158206	20020529
	US 2003235595	A1	20031225	US 2003-397969	20030325
	US 2003236236	A1	20031225	US 2003-444935	20030522
	PRAI	US 1999-345615	A2	19990630	
US 1999-375636		A2	19990817		
US 2000-751968		A2	20001229		
US 1999-258654		A1	19990226		
US 1999-447690		A3	19991123		
WO 2000-US18807		A	20000710		
US 2000-716029		A2	20001117		
US 2001-800593		A2	20010306		
US 2001-877541		A2	20010608		
US 2001-898553		A2	20010702		

AB The present invention relates to pharmaceutical compns. and methods for improved solubilization of triglycerides and improved delivery of therapeutic agents. Compns. of the present invention include a carrier, where the carrier is formed from a combination of a triglyceride and at least 2 surfactants, at least one of which is hydrophilic. Upon dilution with an aqueous medium, the carrier forms a clear, aqueous dispersion of the triglyceride and surfactants. Thus, a formulation contained soybean oil, 80, Tween-20 200, and Tween-80 800 mg.

IT 113665-84-2, Clopidogrel

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
(clear oil-containing pharmaceutical compns. containing therapeutic agent)

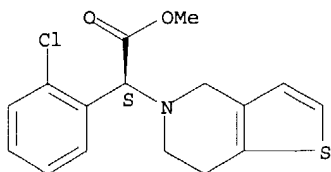
RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-

10686666

dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



L14 ANSWER 14 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2002:185688 CAPLUS

DN 136:252567

TI Methods for drug administration and distribution based on monitoring blood viscosity and other parameters for diagnostics and treatment

IN Kensey, Kenneth

PA USA

SO U.S. Pat. Appl. Publ., 46 pp., Cont.-in-part of U.S. Ser. No. 819,924.

CODEN: USXXCO

DT Patent

LA English

FAN.CNT 7

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2002032149	A1	20020314	US 2001-841389	20010424
	US 6019735	A	20000201	US 1997-919906	19970828
	CA 2301161	AA	19990304	CA 1998-2301161	19980826
	NZ 502905	A	20010831	NZ 1998-502905	19980826
	JP 2001514384	T2	20010911	JP 2000-507994	19980826
	US 6322524	B1	20011127	US 1999-439795	19991112
	US 6322525	B1	20011127	US 2000-501856	20000210
	NO 2000000944	A	20000225	NO 2000-944	20000225
	US 6428488	B1	20020806	US 2000-615340	20000712
	US 2001039828	A1	20011115	US 2001-789350	20010221
	US 2002007664	A1	20020124	US 2001-897164	20010702
	US 6484565	B2	20021126		
	US 2002088953	A1	20020711	US 2001-33841	20011227
	US 6624435	B2	20030923		
	WO 2002079778	A2	20021010	WO 2002-US3984	20020207
	WO 2002079778	A3	20030710		
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW				
	RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
	US 2002184941	A1	20021212	US 2002-156165	20020528
	US 6571608	B2	20030603		
PRAI	US 1997-919906	A2	19970828		
	US 1999-439795	A2	19991112		
	US 2000-501856	A2	20000210		
	US 2000-628401	A2	20000801		
	US 2000-727950	A2	20001201		
	US 2001-819924	A2	20010328		
	US 1997-966076	A	19971107		
	WO 1998-US17657	W	19980826		
	KR 2000-16044	A	20000329		
	US 2000-615340	A3	20000712		
	US 2000-228612P	P	20000828		
	US 2001-789350	A2	20010221		
	US 2001-828761	A	20010409		
	US 2001-839785	A	20010420		
	US 2001-841389	A	20010424		
	US 2001-897164	A3	20010702		
AB	Various methods are provided for determining and utilizing the viscosity of the circulating blood of a living being, i.e., a human, over a range of shear rates for diagnostics and treatment, such as detecting/reducing blood viscosity, work of the heart, contractility of the heart, for				

10686666

detecting/reducing the surface tension of the blood, for detecting plasma viscosity, for explaining/countering endothelial cell dysfunction, for providing high and low blood vessel wall shear stress data, red blood cell deformability data, lubricity of blood, and for treating different ailments such as peripheral arterial disease in combination with administering to a living being at least one pharmaceutically acceptable agent. Agents pharmaceutically effective to regulate at least one of the afore mentioned blood parameters are used to adjust distribution of a substance through the bloodstream. For example, when blood viscosity is a blood flow parameter monitored, an agent is selected from i.v. diluents, red blood cell deformability agents, antiurea agents, oral contraceptives, antidiabetic agents, antiarrhythmics, antihypertensives, antihyperlipidemics, antiplatelet agents, appetite suppressants, antiobesity agents, blood modifiers, smoking deterrent agents, and nutritional supplements.

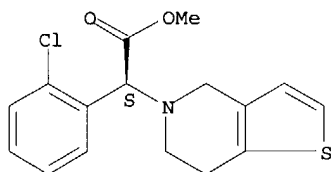
IT 113665-84-2, Clopidogrel

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
(apparatus and methods for monitoring blood viscosity and other parameters in drug delivery for diagnostics and treatment)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



L14 ANSWER 15 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2001:137028 CAPLUS

DN 134:183503

TI Pharmaceutical composition with antithrombotic activity consisting of clopidogrel hydrogen sulfate and a GPIIb/IIIa receptor antagonist

IN Bernat, Andre; Herbert, Jean Marc; Maffrand, Jean Pierre; Savi, Pierre

PA Sanofi-Synthelabo, Fr.

SO PCT Int. Appl., 19 pp.

CODEN: PIXXD2

DT Patent

LA French

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2001012194	A1	20010222	WO 2000-FR2270	20000808
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG				
FR 2797400	A1	20010216	FR 1999-10392	19990811

PRAI FR 1999-10392 A 19990811

AB The invention concerns a combination with antithrombotic activity of two antiaggregating active principles, consisting of clopidogrel hydrogen sulfate (I) and an antagonist of the fibrinogen GPIIb/IIIa receptors (anti-GPIIb/IIIa). The antithrombotic activity of both I and Et 3-[(4-{4-[amino(ethoxycarbonylimino)methyl]phenyl}-1,3-thiazol-2-yl)-(1-ethoxycarbonylmethyl)pyridin-4-yl)amino]propionate (II) was studied in rabbits. A capsule contained I 97.875, II 20.000, pregelatinized starch 40.000, mannitol 233.125, colloidal silica 2.000, and hydrogenated castor oil 7.000 mg.

IT 120202-66-6, Clopidogrel hydrogen sulfate

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(pharmaceutical composition with antithrombotic activity consisting of

10686666

clopidogrel hydrogen sulfate and GPIIb/IIIa receptor antagonist)

RN 120202-66-6 CAPLUS

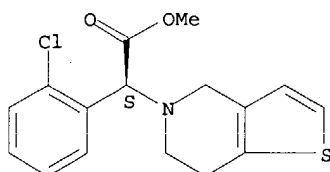
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

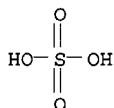
Absolute stereochemistry. Rotation (+).



CM 2

CRN 7664-93-9

CMF H2 O4 S



RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 16 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2000:725436 CAPLUS

DN 133:301171

TI Compositions and methods for improved delivery of ionizable hydrophobic therapeutic agents

IN Chen, Feng-jing; Patel, Manesh V.

PA Lipocine, Inc., USA

SO PCT Int. Appl., 99 pp.

CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2000059475	A1	20001012	WO 2000-US7342	20000316
	W:	AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			
	RW:	GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG			
	US 6383471	B1	20020507	US 1999-287043	19990406
	EP 1165048	A1	20020102	EP 2000-916547	20000316
	R:	AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO			
PRAI	US 1999-287043	A	19990406		
	WO 2000-US7342	W	20000316		

AB The present invention is directed to a pharmaceutical composition including a hydrophobic therapeutic agent having at least one ionizable functional group, and a carrier. The carrier includes an ionizing agent capable of ionizing the functional group, a surfactant, and optionally solubilizers, triglycerides, and neutralizing agents. The invention further relates to a method of preparing such compns. by providing a composition of an ionizable

10686666

hydrophobic therapeutic agent, an ionizing agent, and a surfactant, and neutralizing a portion of the ionizing agent with a neutralizing agent. The compns. of the invention are particularly suitable for use in oral dosage forms. A carrier containing concentrated phosphoric acid 0.025, Tween-20 0.3, Arlacel 186 0.2, sodium taurocholate 0.15, propylene glycol 0.3 g was formulated. Itraconazole was included in the carrier at 30 mg/mL for testing the stability of the itraconazole solution upon dilution in simulated gastric fluid.

IT 113665-84-2, Clopidogrel

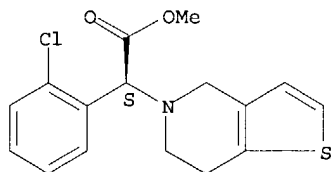
RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(pharmaceutical compns. containing hydrophobic therapeutic agents and carriers containing ionizing agents and surfactants and triglycerides)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 17 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2000:608551 CAPLUS

DN 133:213151

TI Pharmaceutical compositions and methods for improved delivery of hydrophobic therapeutic agents

IN Patel, Manesh V.; Chen, Feng-Jing

PA Lipocine, Inc., USA

SO PCT Int. Appl., 98 pp.

CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 12

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2000050007	A1	20000831	WO 2000-US165	20000105
W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
RW: GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG				
US 6294192	B1	20010925	US 1999-258654	19990226
AU 2000022242	A5	20000914	AU 2000-22242	20000105
AU 771659	B2	20040401		
NZ 513810	A	20010928	NZ 2000-513810	20000105
EP 1158959	A1	20011205	EP 2000-901394	20000105
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO				
JP 2002537317	T2	20021105	JP 2000-600619	20000105
PRAI US 1999-258654	A	19990226		
WO 2000-US165	W	20000105		

AB The present invention relates to triglyceride-free pharmaceutical compns. for delivery of hydrophobic therapeutic agents. Compns. of the present invention include a hydrophobic therapeutic agent and a carrier, where the carrier is formed from a combination of a hydrophilic surfactant and a hydrophobic surfactant. Upon dilution with an aqueous solvent, the composition forms a clear, aqueous dispersion of the surfactants containing the therapeutic agent. The invention also provides methods of treatment with hydrophobic therapeutic agents using these compns. A pharmaceutical composition contained cyclosporin 0.14, Cremophor RH-40 0.41, Arlacel186 0.29, sodium taurocholate 0.26, and propylene glycol 0.46 mg.

IT 113665-84-2, Clopidogrel

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)

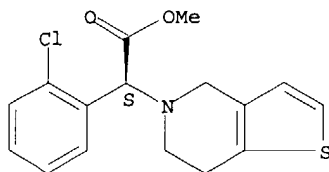
10686666

(pharmaceutical compns. and methods for improved delivery of hydrophobic therapeutic agents)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 18 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2000:144709 CAPLUS

DN 132:185449

TI Pharmaceutical composition for injection based on a pharmaceutically acceptable clopidogrel or ticlopidin salt

IN Aleman, Claude; Breul, Thierry

PA Sanofi-Synthelabo, Fr.

SO PCT Int. Appl., 17 pp.

CODEN: PIXXD2

DT Patent

LA French

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2000010534	A1	20000302	WO 1999-FR2003	19990818
W:	AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			
RW:	GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG			
FR 2782455	A1	20000225	FR 1998-10567	19980820
FR 2782455	B3	20000915		
AU 9951736	A1	20000314	AU 1999-51736	19990818
EP 1105102	A1	20010613	EP 1999-936748	19990818
EP 1105102	B1	20030102		
R:	AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO			
AT 230258	E	20030115	AT 1999-936748	19990818
PT 1105102	T	20030430	PT 1999-936748	19990818
ES 2189456	T3	20030701	ES 1999-936748	19990818
PRAI FR 1998-10567	A	19980820		
WO 1999-FR2003	W	19990818		

AB An aqueous pharmaceutical composition contains a lyophilizate consisting of clopidogrel or ticlopidin optionally in the form of a pharmaceutically acceptable salt, of Pluronic F68 reconstituted in a recovery solvent comprising a basic pH modifying agent compatible with parenteral administration and Solutol HS15. A pharmaceutical lyophilizate contained clopidogrel hydrogen sulfate 100, Pluronic F68 25 mg, and water q.s. 5mL.

IT 120202-66-6, Clopidogrel hydrogen sulfate

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)
(pharmaceutical composition for injection based on pharmaceutically acceptable clopidogrel or ticlopidin salt)

RN 120202-66-6 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate (1:1) (9CI) (CA INDEX NAME)

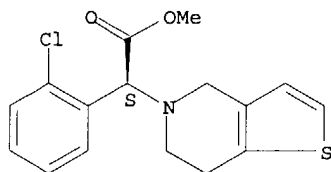
CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

10686666

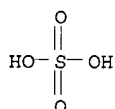
Absolute stereochemistry. Rotation (+).



CM 2

CRN 7664-93-9

CMF H2 O4 S



RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 19 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN
AN 1999:811251 CAPLUS
DN 132:54841
TI Polymorphic clopidogrel hydrogen sulphate form
IN Bousquet, Andre; Castro, Bertrand; Saint-Germain, Jean
PA Sanofi-Synthelabo, Fr.
SO PCT Int. Appl., 32 pp.
CODEN: PIXXD2

DT Patent

LA French

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 9965915	A1	19991223	WO 1999-FR1371	19990610
	W:	AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			
	RW:	GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG			
	FR 2779726	A1	19991217	FR 1998-7464	19980615
	FR 2779726	B1	20010518		
	CA 2334870	AA	19991223	CA 1999-2334870	19990610
	AU 9940483	A1	20000105	AU 1999-40483	19990610
	AU 752170	B2	20020905		
	BR 9911219	A	20010306	BR 1999-11219	19990610
	TR 200003417	T2	20010321	TR 2000-200003417	19990610
	EP 1087976	A1	20010404	EP 1999-923711	19990610
	EP 1087976	B1	20020814		
	R:	AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO			
	EE 200000745	A	20020415	EE 2000-745	19990610
	EE 3972	B1	20030217		
	JP 2002518399	T2	20020625	JP 2000-554740	19990610
	AT 222256	E	20020815	AT 1999-923711	19990610
	NZ 507914	A	20021126	NZ 1999-507914	19990610
	PT 1087976	T	20021129	PT 1999-923711	19990610
	ES 2181439	T3	20030216	ES 1999-923711	19990610
	CN 1128805	B	20031126	CN 1999-807458	19990610
	ZA 2000006386	A	20010507	ZA 2000-6386	20001107
	BG 104987	A	20011130	BG 2000-104987	20001127
	NO 2000006395	A	20010215	NO 2000-6395	20001214
	HR 2000000863	A1	20011031	HR 2000-863	20001214
	HR 20000863	B1	20030430		

10686666

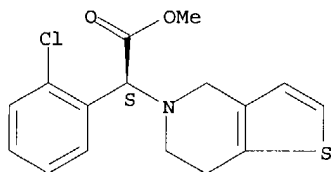
US 6429210 B1 20020806 US 2001-623333 20010405
 HK 1033829 A1 20030328 HK 2001-104337 20010621
 US 2002198229 A1 20021226 US 2002-177092 20020621
 US 6504030 B2 20030107
 PRAI FR 1998-7464 A 19980615
 WO 1999-FR1371 W 19990610
 US 2001-623333 A1 20010405
 AB Novel polymorphic orthorhombic clopidogrel hydrogen **sulfate** or (+)-(S)- α -(2-chlorophenyl)-4,5,6,7-tetrahydrothieno [3,2-c]pyridinyl-5-Me acetate hydrogen **sulfate** form 2 is prepared A solution of 50 g clopidogrel camphosulfonate (preparation given) in 100 mL dichloromethane was added a solution of 9.1 g potassium carbonate in 70 mL water. The organic phase was separated, concentrated, and dissolved in 229 mL acetone. The solution was refluxed with 7.4 g of 80% **sulfuric acid** under N for 2 hj, the solvent was then removed and crystals separated to obtain form 2 clopidogrel hydrogen **sulfate**.
 IT 120202-68-8P
 RL: BPR (Biological process); BSU (Biological study, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); PROC (Process)
 (polymorphic clopidogrel hydrogen **sulfate** form)
 RN 120202-68-8 CAPLUS
 CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, (1R,4S)-7,7-dimethyl-2-oxobicyclo[2.2.1]heptane-1-methanesulfonate (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

Absolute stereochemistry. Rotation (+).

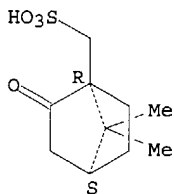


CM 2

CRN 35963-20-3

CMF C10 H16 O4 S

Absolute stereochemistry. Rotation (-).



IT 120202-66-6P, Clopidogrel hydrogen **sulfate**
 RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
 (polymorphic clopidogrel hydrogen **sulfate** form)
 RN 120202-66-6 CAPLUS
 CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate (1:1) (9CI) (CA INDEX NAME)

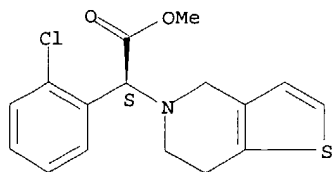
CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

Absolute stereochemistry. Rotation (+).

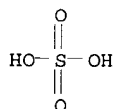
10686666



CM 2

CRN 7664-93-9

CMF H2 O4 S

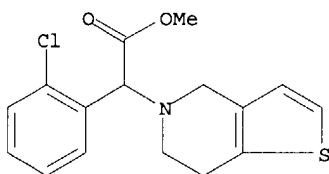


IT 130209-90-4

RL: RCT (Reactant); RACT (Reactant or reagent)
(polymorphic clopidogrel hydrogen **sulfate** form)

RN 130209-90-4 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, hydrochloride (9CI) (CA INDEX NAME)



● HCl

RE.CNT 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 20 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1996:722513 CAPLUS

DN 126:1200

TI Method for the secondary prevention of ischemic events

IN Herbert, Jean-marc; Frehel, Daniel; Bernat, Andre; Badorc, Alain; Savi, Pierre; Delebassee, Denis; Kieffer, Gilles; Defreyn, Ghislain; Maffrand, Jean-pierre

PA Elf Sanofi SA, Fr.

SO U.S., 5 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 5576328	A	19961119	US 1994-190332	19940131
PRAI	US 1994-190332		19940131		

AB The invention relates to a new method for the secondary prevention of ischemic events comprising administering to a man a therapeutically effective amount of a compound selected from clopidogrel and its pharmaceutically acceptable acid addition salts in association with a pharmaceutically acceptable carrier.

IT 113665-84-2, Clopidogrel 120202-66-6, Clopidogrel hydrogen sulfate

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); THU (Therapeutic use); BIOL (Biological study); USES

10686666

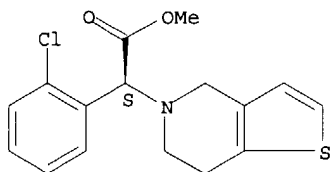
(Uses)

(clopidogrel for secondary prevention of ischemic events)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



RN 120202-66-6 CAPLUS

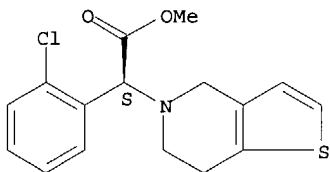
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)-, sulfate (1:1) (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

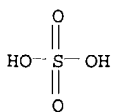
Absolute stereochemistry. Rotation (+).



CM 2

CRN 7664-93-9

CMF H2 O4 S



L14 ANSWER 21 OF 21 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1992:584872 CAPLUS

DN 117:184872

TI Methyl (S)-(chlorophenyl)(tetrahydrothienopyridinyl)acetate for treatment of heart disorders

IN Shibano, Toshiro; Morishima, Yoshuki

PA Daiichi Seiyaku Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 3 pp.

CODEN: JKXXAF

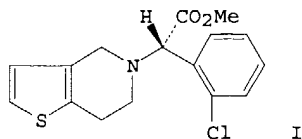
DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 04164033	A2	19920609	JP 1990-291002	19901026
	JP 2949366	B2	19990913		
PRAI	JP 1990-291002		19901026		
GI					

10686666



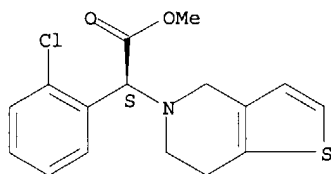
AB Title compound I, or its salts, is useful for prophylactic and therapeutic treatment of heart disorders (e.g. angina pectoris, myocardial infarction, and heart failure). Administration of I **sulfate** at 3 mg/kg i.v. inhibited coronary circulation failure in dogs. LD50 of I **sulfate** was 2603 and 2379 mg/kg p.o. in male and female mice, resp.

IT 113665-84-2 144077-07-6
RL: BIOL (Biological study)
(heart disorders treatment by)

RN 113665-84-2 CAPLUS

CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (α S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



RN 144077-07-6 CAPLUS

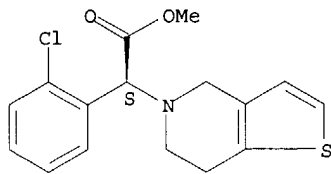
CN Thieno[3,2-c]pyridine-5(4H)-acetic acid, α -(2-chlorophenyl)-6,7-dihydro-, methyl ester, (S)-, sulfate (9CI) (CA INDEX NAME)

CM 1

CRN 113665-84-2

CMF C16 H16 Cl N O2 S

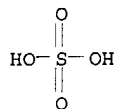
Absolute stereochemistry. Rotation (+).



CM 2

CRN 7664-93-9

CMF H2 O4 S



10686666

=> d his

(FILE 'HOME' ENTERED AT 17:23:59 ON 16 JUN 2004)

FILE 'MEDLINE' ENTERED AT 17:24:21 ON 16 JUN 2004

L1 87 S HYDROGEN SULFATE
L2 56 S ALKYL SULFATE OR (ALKYL SULFATE)
L3 25 S HYDROGEN SULFATE
L4 110 S L1 OR L3
L5 0 S L4 AND L2

FILE 'CAPLUS' ENTERED AT 17:26:00 ON 16 JUN 2004

L6 6805 S HYDROGEN SULFATE OR (HYDROGEN SULFATE)
L7 5050 S (ALKYL SULFATE) OR ALKYL SULFATE
L8 12 S L6 (P) L7
L9 8 S L8 AND SALT?

FILE 'MEDLINE' ENTERED AT 17:34:09 ON 16 JUN 2004

L10 1 S L2 AND PLATELET?
L11 0 S L2 AND THROMBOSIS
L12 6 S L2 AND SALT?

=>

10686666

=> d 1-6 bib abs kwic

L12 ANSWER 1 OF 6 MEDLINE on STN
AN 2002664879 MEDLINE
DN PubMed ID: 12425478
TI Dissolution and partitioning behavior of hydrophobic ion-paired compounds.
AU Lengsfeld C S; Pitera D; Manning M; Randolph T W
CS University of Colorado at Boulder, Department of Chemical Engineering,
Engineering Center, 80309, USA.
SO Pharmaceutical research, (2002 Oct) 19 (10) 1572-6.
Journal code: 8406521. ISSN: 0724-8741.
CY United States
DT Journal; Article; (JOURNAL ARTICLE)
LA English
FS Priority Journals
EM 200304
ED Entered STN: 20021112
Last Updated on STN: 20030425
Entered Medline: 20030424
AB PURPOSE: This study was conducted to determine the effects of counterion hydrophobicity on organic/aqueous partition coefficients for hydrophobic ion paired (HIP) complexes. Furthermore, the coupled dissolution and reverse ion-exchange kinetics for dissolution of HIP complexes into aqueous electrolyte solutions were measured and mathematically modeled. METHODS: HIP complexes of model drugs tacrine and 1-phenylephrine were formed using linear sodium alkylsulfates and bis (2-ethylhexyl sodium sulfosuccinate). Equilibrium partition coefficients between chloroform and aqueous solutions for the complexes and the kinetics of dissolution of the complexes in buffered aqueous solutions were measured. RESULTS: The chloroform/aqueous partition coefficients for 1-phenylephrine/bis (2-ethylhexyl sodium sulfosuccinate) complexes decrease with increasing molar surface tension increment of salts added to the aqueous solution. The logarithm of the partition coefficient for a homologous series of alkyl sulfate complexes decreases as the hydrophilic-lipophilic balance number increases. Dissolution of HIP complexes in deionized water shows first order kinetics, whereas dissolution in aqueous electrolyte solutions shows biphasic kinetics. A kinetic model explains these dissolution rates. CONCLUSIONS: Solubility and dissolution rates for HIP complexes depend on the hydrophobic-lipophilic balance number of the organic counter ion as well as on the electrolyte composition of aqueous solutions. Reverse ion-exchange kinetics are sufficiently slow to allow HIP complexes to be considered simple prodrugs.
AB . . . measured. RESULTS: The chloroform/aqueous partition coefficients for 1-phenylephrine/bis (2-ethylhexyl sodium sulfosuccinate) complexes decrease with increasing molar surface tension increment of salts added to the aqueous solution. The logarithm of the partition coefficient for a homologous series of alkyl sulfate complexes decreases as the hydrophilic-lipophilic balance number increases. Dissolution of HIP complexes in deionized water shows first order kinetics, whereas. . .

L12 ANSWER 2 OF 6 MEDLINE on STN
AN 92272434 MEDLINE
DN PubMed ID: 1590583
TI Liquid chromatographic separation of alkanesulfonate and alkyl sulfate surfactants: effect of ionic strength.
AU Zhou D; Pietrzyk D J
CS University of Iowa, Chemistry Department, Iowa City 52242.
NC 8613
SO Analytical chemistry, (1992 May 1) 64 (9) 1003-8.
Journal code: 0370536. ISSN: 0003-2700.
CY United States
DT Journal; Article; (JOURNAL ARTICLE)
LA English
FS Priority Journals
EM 199206
ED Entered STN: 19920710
Last Updated on STN: 19920710
Entered Medline: 19920622
AB The retention of alkanesulfonate and alkyl sulfate surfactants, which was determined on a reversed stationary phase as a function of mobile-phase ionic strength, is consistent with a double-layer type interaction at the stationary-phase surface. Increasing the mobile-phase ionic strength not only increases retention but also improves resolution because peak widths are significantly reduced. The type of cation provided by the ionic strength salt also enhances retention, reduces peak width, and improves resolution. Lithium hydroxide

is an ideal electrolyte for the separation of multicomponent mixtures of alkanesulfonate and **alkyl sulfate** surfactants. When the column effluent is passed through a postcolumn anion micromembrane suppressor, the conductivity due to the electrolyte is minimized and conductivity detection is sensitive, yielding a detection limit of about 0.3 nmol of injected analyte for a 3:1 signal:noise ratio. Multicomponent alkanesulfonate and **alkyl sulfate** mixtures from C2 to C18 are baseline resolved by using a mobile-phase gradient whereby CH3CN concentration increases and LiOH concentration decreases.

TI Liquid chromatographic separation of alkanesulfonate and **alkyl sulfate** surfactants: effect of ionic strength.

AB The retention of alkanesulfonate and **alkyl sulfate** surfactants, which was determined on a reversed stationary phase as a function of mobile-phase ionic strength, is consistent with a . . . retention but also improves resolution because peak widths are significantly reduced. The type of cation provided by the ionic strength **salt** also enhances retention, reduces peak width, and improves resolution. Lithium hydroxide is an ideal electrolyte for the separation of multicomponent mixtures of alkanesulfonate and **alkyl sulfate** surfactants. When the column effluent is passed through a postcolumn anion micromembrane suppressor, the conductivity due to the electrolyte is . . . sensitive, yielding a detection limit of about 0.3 nmol of injected analyte for a 3:1 signal:noise ratio. Multicomponent alkanesulfonate and **alkyl sulfate** mixtures from C2 to C18 are baseline resolved by using a mobile-phase gradient whereby CH3CN concentration increases and LiOH concentration. . . .

L12 ANSWER 3 OF 6 MEDLINE on STN

AN 89340739 MEDLINE

DN PubMed ID: 2547806

TI Separation and indirect detection of alkyl sulfonates and sulfates.

AU Pietrzyk D J; Rigas P G; Yuan D X

CS University of Iowa, Chemistry Department, Iowa City 52240.

NC DE 8613 (NIDCR)

SO Journal of chromatographic science, (1989 Aug) 27 (8) 485-90.

Journal code: 0173225. ISSN: 0021-9665.

CY United States

DT Journal; Article; (JOURNAL ARTICLE)

LA English

FS Priority Journals

EM 198909

ED Entered STN: 19900309

Last Updated on STN: 20000303

Entered Medline: 19890921

AB Iron(II) 1,10-phenanthroline, Fe(phen)3(2+), **salts** are used as mobile phase additives for the liquid chromatographic separation of alkyl sulfonates and sulfates on the reversed-phase PRP-1. As alkyl chain length increases retention increases. For a given chain length an **alkyl sulfate** is more retained than the corresponding alkyl sulfonate. Major elution variables that affect retention are mobile phase solvent and counteranion concentration. Indirect photometric detection is used to detect alkyl sulfonates and sulfates at 510 nm where Fe(phen)3(2+) **salts** absorb. Conditions for isocratic and gradient elution of multicomponent mixtures are described. Detection limits depending on analyte approached 0.1 nmol for isocratic elution and 3 nmol for gradient elution.

AB Iron(II) 1,10-phenanthroline, Fe(phen)3(2+), **salts** are used as mobile phase additives for the liquid chromatographic separation of alkyl sulfonates and sulfates on the reversed-phase PRP-1. As alkyl chain length increases retention increases. For a given chain length an **alkyl sulfate** is more retained than the corresponding alkyl sulfonate. Major elution variables that affect retention are mobile phase solvent and counteranion concentration. Indirect photometric detection is used to detect alkyl sulfonates and sulfates at 510 nm where Fe(phen)3(2+) **salts** absorb. Conditions for isocratic and gradient elution of multicomponent mixtures are described. Detection limits depending on analyte approached 0.1 nmol. . . .

L12 ANSWER 4 OF 6 MEDLINE on STN

AN 62115651 MEDLINE

DN PubMed ID: 13924023

TI Studies on surface activation of medicinals. VI. **Salt** formation of **alkylsulfate** of various amino compounds. (2). On various basic medicines.

AU UTSUMI I; HARADA K

SO Japanese journal of pharmacology, (1962 Jan) 82 108-14.

Journal code: 2983305R. ISSN: 0021-5198.

DT Journal; Article; (JOURNAL ARTICLE)

10686666

LA Japanese
FS OLDMEDLINE
EM 199811
ED Entered STN: 19990716
Last Updated on STN: 19990716
Entered Medline: 19981101
TI Studies on surface activation of medicinals. VI. **Salt** formation
of **alkylsulfate** of various amino compounds. (2). On various
basic medicines.

L12 ANSWER 5 OF 6 MEDLINE on STN
AN 62115650 MEDLINE
DN PubMed ID: 13924022
TI Studies on surface activation of medicinals. V. **Salt** formations
of **alkylsulfate** of various amino compounds. (1). On the
alkyl-amines and amino acids.
AU UTSUMI I; HARADA K
SO Japanese journal of pharmacology, (1962 Jan) 82 102-7.
Journal code: 2983305R. ISSN: 0021-5198.
DT Journal; Article; (JOURNAL ARTICLE)
LA Japanese
FS OLDMEDLINE
EM 199811
ED Entered STN: 19990716
Last Updated on STN: 19990716
Entered Medline: 19981101
TI Studies on surface activation of medicinals. V. **Salt** formations
of **alkylsulfate** of various amino compounds. (1). On the
alkyl-amines and amino acids.

L12 ANSWER 6 OF 6 MEDLINE on STN
AN 60225925 MEDLINE
DN PubMed ID: 13854943
TI Esters of erythromycin. IV. **Alkyl sulfate**
salts.
AU STEPHENS V C; CONINE J W; MURPHY H W
SO Journal of the American Pharmaceutical Association. American
Pharmaceutical Association, (1959 Nov) 48 620-2.
Journal code: 14840180R.
DT Journal; Article; (JOURNAL ARTICLE)
LA English
FS OLDMEDLINE
EM 199811
ED Entered STN: 19990716
Last Updated on STN: 19990716
Entered Medline: 19981101
TI Esters of erythromycin. IV. **Alkyl sulfate**
salts.

10686666

=> d his

(FILE 'HOME' ENTERED AT 17:23:59 ON 16 JUN 2004)

FILE 'MEDLINE' ENTERED AT 17:24:21 ON 16 JUN 2004

L1 87 S HYDROGEN SULFATE
L2 56 S ALKYL SULFATE OR (ALKYL SULFATE)
L3 25 S HYDROGEN SULFATE
L4 110 S L1 OR L3
L5 0 S L4 AND L2

FILE 'CAPLUS' ENTERED AT 17:26:00 ON 16 JUN 2004

L6 6805 S HYDROGEN SULFATE OR (HYDROGEN SULFATE)
L7 5050 S (ALKYL SULFATE) OR ALKYL SULFATE
L8 12 S L6 (P) L7
L9 8 S L8 AND SALT?

=>

10686666

=> d 1-8 bib abs kwic

L9 ANSWER 1 OF 8 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2003:397011 CAPLUS
DN 138:398091
TI Actinomycetes secondary alkylsulfatases and their use for enantioselective
hydrolysis of secondary alkylsulfate esters
IN Faber, Kurt; Pogorevc, Mateja; Riermeier, Thomas
PA Degussa A.-G., Germany
SO PCT Int. Appl., 48 pp.
CODEN: PIXXD2
DT Patent
LA German
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2003042378	A1	20030522	WO 2002-EP12618	20021112
	W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			
	RW:	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG			

DE 10155764 A1 20030528 DE 2001-10155764 20011114
PRAI DE 2001-10155764 A 20011114

AB The invention concerns novel alkylsulfatases obtained from Actinomycetes, especially Rhodococcus, and the use of the enzymes for enantioselective hydrolysis of secondary alkylsulfate esters to produce chiral secondary alcs. Thus, two alkylsulfatases were purified from Rhodococcus ruber and characterized. One of the enzymes had a pH optimum between 7.5 and 8.0 and a temperature optimum around 30° with 2-octylsulfate as substrate. C7-10-alkylsulfates were preferred substrates for this enzyme. Inclusion of ferrrous or ferric salts or CMAB in the reaction mixture increased the enantioselectivity of the reaction.

RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

AB The invention concerns novel alkylsulfatases obtained from Actinomycetes, especially Rhodococcus, and the use of the enzymes for enantioselective hydrolysis of secondary alkylsulfate esters to produce chiral secondary alcs. Thus, two alkylsulfatases were purified from Rhodococcus ruber and characterized. One of the enzymes had a pH optimum between 7.5 and 8.0 and a temperature optimum around 30° with 2-octylsulfate as substrate. C7-10-alkylsulfates were preferred substrates for this enzyme. Inclusion of ferrrous or ferric salts or CMAB in the reaction mixture increased the enantioselectivity of the reaction.

IT 34760-88-8, 2-Octylsulfate 74403-66-0, 4-Octanol, hydrogen sulfate 74403-67-1, 3-Octanol, hydrogen sulfate 103142-09-2, 2-Nonylsulfate
RL: RCT (Reactant); RACT (Reactant or reagent)
(hydrolysis of; actinomycetes secondary alkylsulfatases and their use for enantioselective hydrolysis of secondary alkylsulfate esters)

IT 57-09-0, Cetyltrimethylammonium bromide 7705-08-0, Ferric chloride, biological studies 7758-94-3, Ferrous chloride 15438-31-0D, Iron(2+), salts, biological studies 20074-52-6D, Iron(3+), salts, biological studies
RL: BUU (Biological use, unclassified); BIOL (Biological study); USES (Uses)
(in enzymic hydrolysis of sulfate esters; actinomycetes secondary alkylsulfatases and their use for enantioselective hydrolysis of secondary alkylsulfate esters)

L9 ANSWER 2 OF 8 CAPLUS COPYRIGHT 2004 ACS on STN
AN 2002:850150 CAPLUS
DN 137:346926
TI Alkyl sulfate salts as male contraceptives
IN Zimmerman, Ronald
PA USA
SO U.S. Pat. Appl. Publ., 13 pp.
CODEN: USXXCO
DT Patent
LA English
FAN.CNT 1

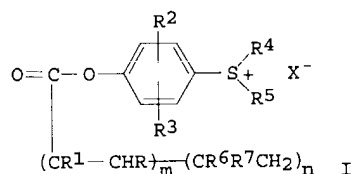
10686666

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2002164368	A1	20021107	US 2001-22671	20011217
PRAI	US 2000-256370P	P	20001218		
OS	MARPAT 137:346926				
AB	The present invention relates to a male contraceptive composition suitable for oral administration, wherein the composition comprises a pharmaceutically acceptable, non-toxic cationic salt of an alkyl sulfate. For example, tetradecyl sodium sulfate administered orally at a dose of 10 mg/kg to male rabbits inhibited the fertilization. Tetradecyl sodium sulfate was observed to bind to the entire sperm plasma membranes and not just to acrosomes.				
TI	Alkyl sulfate salts as male contraceptives				
AB	The present invention relates to a male contraceptive composition suitable for oral administration, wherein the composition comprises a pharmaceutically acceptable, non-toxic cationic salt of an alkyl sulfate. For example, tetradecyl sodium sulfate administered orally at a dose of 10 mg/kg to male rabbits inhibited the fertilization. Tetradecyl sodium sulfate was observed to bind to the entire sperm plasma membranes and not just to acrosomes.				
ST	alkyl sulfate cationic salt oral male contraceptive				
IT	Sperm (binding to plasma membrane of; alkyl sulfate salts as oral male contraceptives)				
IT	Drug delivery systems (capsules, soft; alkyl sulfate salts as oral male contraceptives)				
IT	Drug delivery systems (capsules; alkyl sulfate salts as oral male contraceptives)				
IT	Drug delivery systems (injections, s.c.; alkyl sulfate salts as oral male contraceptives)				
IT	Contraceptives (male, oral; alkyl sulfate salts as oral male contraceptives)				
IT	Contraceptives (oral, male; alkyl sulfate salts as oral male contraceptives)				
IT	Drug delivery systems (suspensions, oral; alkyl sulfate salts as oral male contraceptives)				
IT	Drug delivery systems (tablets; alkyl sulfate salts as oral male contraceptives)				
IT	Drug delivery systems (transdermal; alkyl sulfate salts as oral male contraceptives)				
IT	112-03-8 139-96-8 142-78-9 142-87-0, Sodium n-decyl sulfate 143-00-0 1072-24-8, Sodium n-undecyl sulfate 1119-97-7 1120-01-0, Sodium n-hexadecyl sulfate 1120-04-3 1191-50-0, Sodium n-tetradecyl sulfate 1241-94-7 1847-55-8, Sodium oleyl sulfate 2627-35-2 2673-22-5 2958-09-0 3006-15-3 3026-63-9 3700-67-2 3921-30-0 4671-75-4 4721-24-8 4724-48-5 5137-70-2 5910-79-2, Sodium n-heptadecyl sulfate 6482-41-3, Sodium 2-tetradecyl sulfate 6858-55-5, 2-Octadecanol, hydrogen sulfate, sodium salt 6874-60-8 6920-63-4, Sodium 8-hexadecyl sulfate 6920-74-7, Sodium 7-hexadecyl sulfate 9004-98-2 10054-29-2 13177-49-6 13177-50-9 13393-71-0 13419-37-9 14167-87-4 15724-25-1 17006-05-2 18695-78-8 25446-91-7 29454-05-5 36873-80-0 52304-21-9 52886-14-3 59378-31-3 68105-02-2 69214-95-5 71215-57-1 71317-49-2 78204-48-5 78204-49-6 78204-50-9, 6-Tetradecanol, hydrogen sulfate, sodium salt 78204-51-0, 7-Tetradecanol, hydrogen sulfate, sodium salt 78204-53-2, 9-Octadecanol, hydrogen sulfate, sodium salt 78204-55-4 78204-56-5 78204-57-6 78204-58-7 79395-72-5 109727-48-2 119159-04-5 142474-86-0 338734-65-9 338734-66-0 474498-34-5 474498-37-8 474498-38-9 474498-39-0 474498-40-3 474534-59-3 RL: PAC (Pharmacological activity); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (alkyl sulfate salts as oral male contraceptives)				
IT	9068-57-9, Acrosin 37326-33-3, Hyaluronidase RL: BSU (Biological study, unclassified); BIOL (Biological study) (inhibition of; alkyl sulfate salts as oral male contraceptives)				
L9	ANSWER 3 OF 8 CAPLUS COPYRIGHT 2004 ACS on STN				
AN	1993:562460 CAPLUS				
DN	119:162460				
TI	Sulfonium salt group-containing polymers and their use as				

10686666

additives in coatings
 IN Muraoka, Tokuyuki; Takashita, Katsushige; Akashi, Sumio; Koizumi, Tatsuya;
 Nagai, Katsutoshi
 PA Sanshin Kagaku Kogyo Kk, Japan
 SO Jpn. Kokai Tokkyo Koho, 4 pp.
 CODEN: JKXXAF
 DT Patent
 LA Japanese
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 05070720	A2	19930323	JP 1991-308613	19910911
PRAI	JP 1991-308613		19910911		
GI					



AB Title polymers I (R = H, C1-4 alkyl, Ph; R1 = H, C1-4 alkyl; R2, R3 = H, halo, C1-4 alkyl, C1-4 alkoxy; R4, R5 = C1-4 alkyl; X = alkylsulfate, halo, perchlorate, hydrogen sulfate, p-toluenesulfonate; R6, R7 = H, halo, organic group; 0 < m ≤ 100; 0 ≤ n < 100; m + n = 100) are added in 0.1-20.0% proportion to coatings as thickeners, dispersants, and corrosion inhibitors. Thus, a glass bead with 0.5 mm diameter fell 35 cm in 18.21 s through a water-based acrylic paint containing 5% 4-dimethylsulfoniophenyl methacrylate methylsulfate-styrene copolymer (II) vs. 15.74 s in the absence of II.

TI Sulfonium salt group-containing polymers and their use as additives in coatings

AB Title polymers I (R = H, C1-4 alkyl, Ph; R1 = H, C1-4 alkyl; R2, R3 = H, halo, C1-4 alkyl, C1-4 alkoxy; R4, R5 = C1-4 alkyl; X = alkylsulfate, halo, perchlorate, hydrogen sulfate, p-toluenesulfonate; R6, R7 = H, halo, organic group; 0 < m ≤ 100; 0 ≤ n < 100; m + n = 100) are added in 0.1-20.0% proportion to coatings as thickeners, dispersants, and corrosion inhibitors. Thus, a glass bead with 0.5 mm diameter fell 35 cm in 18.21 s through a water-based acrylic paint containing 5% 4-dimethylsulfoniophenyl methacrylate methylsulfate-styrene copolymer (II) vs. 15.74 s in the absence of II.

ST sulfoniophenyl acrylate salt copolymer thickener; dispersant
 sulfoniophenyl acrylate salt copolymer; corrosion inhibitor
 sulfoniophenyl acrylate copolymer

L9 ANSWER 4 OF 8 CAPLUS COPYRIGHT 2004 ACS on STN
 AN 1960:11069 CAPLUS
 DN 54:11069
 OREF 54:2183a-b
 TI Guanidine and its derivatives
 IN Roberts, Elwyn
 PA Minister of Supply, London
 DT Patent
 LA Unavailable
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2884437		19590428	US	
AB	A process for the com. feasible preparation of guanidine and derivs. of guanidine, e.g., nitroguanidine, which is an important flashless high explosive and useful projectile propellant, consists of treating urea with a dialkyl sulfate to produce an alkyl isourea alkyl hydrogen sulfate then treating the alkyl isourea alkyl hydrogen sulfate to produce guanidinium alkyl sulfate (I), treating this with an alkali or alkali alcoholate to produce guanidine (II), and treating this with nitric and sulfuric acids to produce nitroguanidine (III). Hydrolyzing the I to guanidinium hydrogen sulfate, nitrating the guanidinium hydrogen sulfate with nitric and sulfuric acids, to				

intermediate

produce III and treating the guanidinium **alkyl sulfate** with an alkali alcoholate to produce an alcoholic solution of guanidine, and neutralizing the solution with an acid produced the corresponding guanidine **salt**.

AB A process for the com. feasible preparation of guanidine and derivs. of guanidine, e.g., nitroguanidine, which is an important flashless high explosive and useful projectile propellant, consists of treating urea with a dialkyl sulfate to produce an alkyl isourea alkyl **hydrogen sulfate** then treating the alkyl isourea alkyl **hydrogen sulfate** to produce guanidinium **alkyl sulfate** (I), treating this with an alkali or alkali alcoholate to produce guanidine (II), and treating this with nitric and sulfuric acids to produce nitroguanidine (III). Hydrolyzing the I to guanidinium **hydrogen sulfate**, nitrating the guanidinium **hydrogen sulfate** with nitric and sulfuric acids, to produce III and treating the guanidinium **alkyl sulfate** with an alkali alcoholate to produce an alcoholic solution of guanidine, and neutralizing the solution with an acid produced the corresponding guanidine **salt**.

L9 ANSWER 5 OF 8 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1960:10977 CAPLUS

DN 54:10977

OREF 54:2167f-i

TI (Alkylthio)alkyl sulfates

IN Doerr, Edward L.

PA Monsanto Chemical Co.

DT Patent

LA Unavailable

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 2909554		19591020	US	
AB	(Alkylthio)alkyl sulfate salts (I), useful as surface-active agents, nematocides, and fungicides, were prepared by the reaction of 2-(alkylthio)ethanols (II) (8-18 C in the alkyl radical) with chlorosulfonic acid (III) which gave the II hydrogen sulfate , which was neutralized with an alkali metal hydroxide or NH ₄ OH to obtain I. Thus, 34.2 g. III was added dropwise to 60 ml. Et ₂ O with ice-cooling. The mixture was added (9 min.) at 2-5° to 61.6 g. n-dodecylthioethanol in about 500 ml. Et ₂ O, then stirred for 30 min. while in an ice bath, and the Et ₂ O removed in vacuo. The residue was neutralized with 50% NaOH (aqueous-EtOH), excess EtOH added, the mixture heated to 50° and filtered, the filtrate cooled, the crystallized solid separated, dried in vacuo at room temperature, and treated with Me ₂ CO in a mixer. The separated solid was washed with Me ₂ CO and dried in vacuo at room temperature to obtain Na dodecylthioethyl sulfate. Similarly, 2-(tert-dodecylthio)ethanol was sulfated, the sulfate was neutralized with NaOH in aqueous iso-PrOH, and Na 2-(tert-dodecylthio)ethyl sulfate was obtained. The sulfation of 2-(tert-hexadecylthio)ethanol and neutralization of the hydrogen sulfate in EtOH (slight excess of NaOH) yielded Na 2-(tert-hexadecylthio)ethyl sulfate.				
AB	(Alkylthio)alkyl sulfate salts (I), useful as surface-active agents, nematocides, and fungicides, were prepared by the reaction of 2-(alkylthio)ethanols (II) (8-18 C in the alkyl radical) with chlorosulfonic acid (III) which gave the II hydrogen sulfate , which was neutralized with an alkali metal hydroxide or NH ₄ OH to obtain I. Thus, 34.2 g. III was added dropwise to 60 ml. Et ₂ O with ice-cooling. The mixture was added (9 min.) at 2-5° to 61.6 g. n-dodecylthioethanol in about 500 ml. Et ₂ O, then stirred for 30 min. while in an ice bath, and the Et ₂ O removed in vacuo. The residue was neutralized with 50% NaOH (aqueous-EtOH), excess EtOH added, the mixture heated to 50° and filtered, the filtrate cooled, the crystallized solid separated, dried in vacuo at room temperature, and treated with Me ₂ CO in a mixer. The separated solid was washed with Me ₂ CO and dried in vacuo at room temperature to obtain Na dodecylthioethyl sulfate. Similarly, 2-(tert-dodecylthio)ethanol was sulfated, the sulfate was neutralized with NaOH in aqueous iso-PrOH, and Na 2-(tert-dodecylthio)ethyl sulfate was obtained. The sulfation of 2-(tert-hexadecylthio)ethanol and neutralization of the hydrogen sulfate in EtOH (slight excess of NaOH) yielded Na 2-(tert-hexadecylthio)ethyl sulfate.				
IT	7664-93-9,		Sulfuric acid		
	((alkylthio)alkyl esters, Na salts)				
IT	56949-83-8,		Ethanol, 2-(dodecylthio)-,	sulfate, Na salt	
	107308-78-1,		Ethanol, 2-(tert-dodecylthio)-,	sulfate, Na salt	
	111411-10-0,		Ethanol, 2-(tert-hexadecylthio)-,	sulfate, Na salt	
	(preparation of)				

L9 ANSWER 6 OF 8 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1923:5483 CAPLUS

DN 17:5483

OREF 17:992g-i,993a-d

TI Esters of the hydroxyalkylarylamines. I. Acid sulfuric esters of the simple monohydroxyethylarylamines

AU Saunders, K. H.

SO Journal of the Chemical Society, Abstracts (1922), 121, 2667-75

CODEN: JCSAAZ; ISSN: 0590-9791

DT Journal

LA Unavailable

AB The alkylsulfuric acid group attached to N is termed the "sulfato" group, and the process of esterification as "sulfation." These esters may be prepared in 3 ways: By solution in concentrated H₂SO₄ in such excess that esterification proceeds virtually to completion. This process is beset with the same difficulties found in the attempt to isolate EtHSO₄ itself in high yield and purity. A 2nd method consists in acting on arylamines with ClCH₂CH₂OSO₃H, which has the disadvantage that the latter must be prepared from anhydrous HOC₂H₄Cl and the yields are not always good. The 3rd method consists in esterifying with ClSO₃H which may be used alone or in an indifferent solvent. In this reaction the neutral ester is also obtained in varying amts. Chemically these sulfato compds. show reactions characteristic of the units of their structure-arylamine and **alkyl sulfate**. N-Phenyl-β-aminoethyl **hydrogen sulfate** (sulfaloethylaniline), prepared by each of the above 3 methods, rectangular laminas, m. 206°; it decolorizes Br-H₂O but does not react with CuSO₄. It is soluble to the extent of 5% in boiling and 1-2% in cold EtOH. It is very slowly hydrolyzed by H₂O below its b. p.; HCl accelerates the hydrolysis, which has a value of k for a monomol. reaction. Practically no hydrolysis was found after heating with 0.2 or 0.8 N NaOH at 70° for 5 hrs.; heated with 3 mols. NaOH for 1 hr., 20.4% of the **salt** had hydrolyzed. Sodium **salt**, with 1 H₂O, leaflets, soluble to the extent of 60 g. per 100 cc. of solution at 15°; potassium **salt**, leaflets; 23 parts dissolve in 100 cc. H₂O at 15°; ammonium **salt**, leaflets, m. 132°, of which 70 g. dissolve in 100 cc. H₂O at 15°. N-o-Tolyl-β-aminoethyl **hydrogen sulfate**, rectangular laminas, m. 203°. N-Phenyl-N-ethyl-β-aminoethyl **hydrogen sulfate**, hard granules, m. 208°. Treated in N NaOH with solid NaNO₂ after which concentrated HCl was slowly added, this gave the p-nitroso derivative, dark green dust, decomposing 170-80°, readily reduced in alkaline solution N-Phenyl-N-methyl-β-aminoethyl **hydrogen sulfate**, m. 193°. Sodium N-phenyl-N-benzyl-β-aminoethyl sulfate, shining crystals with 2H₂O, which it loses at 100° and then m. to a waxy mass; the free acid could not be obtained crystalline N-m-Nitrophenyl-β-aminoethyl **hydrogen sulfate**, stout pale cream needles, m. 203° (decomposition). The alkaline solution is a deep orange. m-Nitroaniline **salt**, large pale yellow laminas, m. 206°. N-p-Chlorophenyl-β-aminoethyl **hydrogen sulfate**, needles, m. 217° (decomposition). N-α-Naphthyl derivative, m. 234° (decomposition). The coupling with diazo **salts** in the p-position to the sulfato group will be described in a later article. See also Brit. patent 181,750 of 1922.

AB The alkylsulfuric acid group attached to N is termed the "sulfato" group, and the process of esterification as "sulfation." These esters may be prepared in 3 ways: By solution in concentrated H₂SO₄ in such excess that esterification proceeds virtually to completion. This process is beset with the same difficulties found in the attempt to isolate EtHSO₄ itself in high yield and purity. A 2nd method consists in acting on arylamines with ClCH₂CH₂OSO₃H, which has the disadvantage that the latter must be prepared from anhydrous HOC₂H₄Cl and the yields are not always good. The 3rd method consists in esterifying with ClSO₃H which may be used alone or in an indifferent solvent. In this reaction the neutral ester is also obtained in varying amts. Chemically these sulfato compds. show reactions characteristic of the units of their structure-arylamine and **alkyl sulfate**. N-Phenyl-β-aminoethyl **hydrogen sulfate** (sulfaloethylaniline), prepared by each of the above 3 methods, rectangular laminas, m. 206°; it decolorizes Br-H₂O but does not react with CuSO₄. It is soluble to the extent of 5% in boiling and 1-2% in cold EtOH. It is very slowly hydrolyzed by H₂O below its b. p.; HCl accelerates the hydrolysis, which has a value of k for a monomol. reaction. Practically no hydrolysis was found after heating with 0.2 or 0.8 N NaOH at 70° for 5 hrs.; heated with 3 mols. NaOH for 1 hr., 20.4% of the **salt** had hydrolyzed. Sodium **salt**, with 1 H₂O, leaflets, soluble to the extent of 60 g. per 100 cc. of solution at 15°; potassium **salt**, leaflets; 23 parts dissolve in 100 cc. H₂O at 15°; ammonium **salt**, leaflets, m. 132°;

of which 70 g. dissolve in 100 cc. H₂O at 15°.
 N-o-Tolyl-β-aminoethyl **hydrogen sulfate**,
 rectangular laminas, m. 203°. N-Phenyl-N-ethyl-β-aminoethyl
hydrogen sulfate, hard granules, m. 208°.
 Treated in N NaOH with solid NaNO₂ after which concentrated HCl was slowly
 added, this gave the p-nitroso derivative, dark green dust, decomposing
 170-80°, readily reduced in alkaline solution N-Phenyl-N-methyl-β-
 aminoethyl **hydrogen sulfate**, m. 193°. Sodium
 N-phenyl-N-benzyl-β-aminoethyl sulfate, shining crystals with 2H₂O,
 which it loses at 100° and then m. to a waxy mass; the free acid
 could not be obtained crystalline N-m-Nitrophenyl-β-aminoethyl
hydrogen sulfate, stout pale cream needles, m.
 203° (decomposition). The alkaline solution is a deep orange. m-Nitroaniline
salt, large pale yellow laminas, m. 206°.
 N-p-Chlorophenyl-β-aminoethyl **hydrogen sulfate**,
 needles, m. 217° (decomposition). N-α-Naphthyl derivative, m.
 234° (decomposition). The coupling with diazo **salts** in the
 p-position to the sulfato group will be described in a later article. See
 also Brit. patent 181,750 of 1922.

- IT Aniline, sulfatoethyl-
 Ethylsulfuric acid, β-(N-benzylanilino)-, sodium **salt**
 Ethylsulfuric acid, β-(N-ethyl-p-nitrosoanilino)-
 Ethylsulfuric acid, β-(N-ethylanilino)-
 Ethylsulfuric acid, β-(N-methylanilino)-
 Ethylsulfuric acid, β-(m-nitroanilino)-
 Ethylsulfuric acid, β-(m-nitroanilino)-, m-nitroanilino **salt**
 Ethylsulfuric acid, β-(p-chloroanilino)-
 Ethylsulfuric acid, β-o-toluino-
 IT Ethylsulfuric acid, β-anilino-
 (and **salts**)

L9 ANSWER 7 OF 8 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1918:12868 CAPLUS

DN 12:12868

OREF 12:2195c-h

TI Hydrolysis of methyl sulfate and ethyl sulfate with sodium methoxide or
 ethoxide

AU Pollak, J.; Baar, A.

SO Journal of the Chemical Society, Abstracts (1918), 114(II), 361

CODEN: JCSAAZ; ISSN: 0590-9791

DT Journal

LA Unavailable

AB Me₂SO₄ undergoes hydrolysis by H₂O more rapidly than Et₂SO₄, but in the
 presence of KOH the ratio of the reaction velocities is very different
 from that observed for the hydrolysis by H₂O only. With 0.5 N KOH at
 25°, the unimol. constant for MeOH is 45 times as great as for the
 Et₂SO₄, whereas with H₂O only the ratio is approx. 5:1. In order to
 decide whether the difference is due to the difference in the solubility of the
 2 esters in H₂O, and to avoid the possibility of such a disturbing factor,
 it is desirable to exam. the rate of reaction in a homogeneous system.
 Kremann has already observed that with MeOH and EtOH, the rate of reaction
 of reaction of Me₂SO₄ is 3-4 times that of Et₂SO₄. With an alc. solution of
 EtONa, however, at 25° Me₂SO₄ reacts approx. 25 times as rapidly as
 Et₂SO₄, while at 0° the ratio is 58 to 1. The reaction in each
 case proceeds as far as the corresponding alkyl **hydrogen**
sulfate or its Na **salt**, any further hydrolysis being
 negligible. These results demonstrate that the great difference in the
 velocities of reaction of alkali on the two alkyl sulfates is not mainly
 due to any difference of solubility on the part of the sulfates, because a
 similar difference is observed in homogeneous and heterogeneous system.
 The difference is, therefore, presumably to be attributed to the different
 character of the reactions, the alkali hydrolysis yielding the alkali
salt of the alkylsulfuric acid and while the free alkylsulfuric acid is
 produced by the action of H₂O or of alc. Examination of the reaction velocity
 of Et₂SO₄ and Me₂SO₄ with alc. in the presence of a gradually increasing
 portion of water shows that the former ester is distinctly less soluble in
 H₂O and that the difference in the solubility of the 2 esters may exert an
 appreciable influence on the relative apparent activity of the two esters
 towards alkali hydroxide in the heterogeneous system. MeOH reacts with
 the 2 alkyl sulfates more rapidly than does EtOH, and although it was
 found that, as expected, NaOMe affects the Me ester much more rapidly than
 the Et ester. the surprising result was obtained that MeONa in MeOH is
 less reactive than an EtOH solution of EtONa. An observation similar to this
 has already been made in certain cases. The suggestion of Kremann that
 the difference is due to the presence of traces of H₂O which causes a
 greater proportion of hydrolysis in the NaOEt is discredited and the
 suggestion is made that the explanation may be found in the possible
 occurrence of the reaction between the alkyl **sulfate**

and the undissociated portion of the Na alkoxide.M

AB Me2SO4 undergoes hydrolysis by H2O more rapidly than Et2SO4, but in the presence of KOH the ratio of the reaction velocities is very different from that observed for the hydrolysis by H2O only. With 0.5 N KOH at 25°, the unimol. constant for MeOH is 45 times as great as for the Et2SO4, whereas with H2O only the ratio is approx. 5:1. In order to decide whether the difference is due to the difference in the solubility of the 2 esters in H2O, and to avoid the possibility of such a disturbing factor, it is desirable to exam. the rate of reaction in a homogeneous system. Kremann has already observed that with MeOH and EtOH, the rate of reaction of reaction of Me2SO4 is 3-4 times that of Et2SO4. With an alc. solution of EtONa, however, at 25° Me2SO4 reacts approx. 25 times as rapidly as Et2SO4, while at 0° the ratio is 58 to 1. The reaction in each case proceeds as far as the corresponding alkyl **hydrogen sulfate** or its Na salt, any further hydrolysis being negligible. These results demonstrate that the great difference in the velocities of reaction of alkali on the two alkyl sulfates is not mainly due to any difference of solubility on the part of the sulfates, because a similar difference is observed in homogeneous and heterogeneous system. The difference is, therefore, presumably to be attributed to the different character of the reactions, the alkali hydrolysis yielding the alkali **salt** of the alkylsulfuric and while the free alkylsulfuric acid is produced by the action of H2O or of alc. Examination of the reaction velocity of Et2SO4 and Me2SO4 with alc. in the presence of a gradually increasing portion of water shows that the former ester is distinctly less soluble in H2O and that the difference in the solubility of the 2 esters may exert an appreciable influence on the relative apparent activity of the two esters towards alkali hydroxide in the heterogeneous system. MeOH reacts with the 2 alkyl sulfates more rapidly than does EtOH, and although it was found that, as expected, NaOMe affects the Me ester much more rapidly than the Et ester. the surprising result was obtained that MeONa in MeOH is less reactive than an EtOH solution of EtONa. An observation similar to this has already been made in certain cases. The suggestion of Kremann that the difference is due to the presence of traces of H2O which causes a greater proportion of hydrolysis in the NaOEt is discredited and the suggestion is made that the explanation may be found in the possible occurrence of the reaction between the **alkyl sulfate** and the undissociated portion of the Na alkoxide.M

L9 ANSWER 8 OF 8 CAPLUS COPYRIGHT 2004 ACS on STN

AN 1918:12867 CAPLUS

DN 12:12867

OREF 12:2195c-h

TI Hydrolysis of methyl sulfate and ethyl sulfate with sodium methoxide or ethoxide

AU Pollak, J.; Baar, A.

SO Monatshefte fuer Chemie (1918), 38, 501-23

CODEN: MOCMB7; ISSN: 0026-9247

DT Journal

LA Unavailable

AB Me2SO4 undergoes hydrolysis by H2O more rapidly than Et2SO4, but in the presence of KOH the ratio of the reaction velocities is very different from that observed for the hydrolysis by H2O only. With 0.5 N KOH at 25°, the unimol. constant for MeOH is 45 times as great as for the Et2SO4, whereas with H2O only the ratio is approx. 5:1. In order to decide whether the difference is due to the difference in the solubility of the 2 esters in H2O, and to avoid the possibility of such a disturbing factor, it is desirable to exam. the rate of reaction in a homogeneous system. Kremann has already observed that with MeOH and EtOH, the rate of reaction of reaction of Me2SO4 is 3-4 times that of Et2SO4. With an alc. solution of EtONa, however, at 25° Me2SO4 reacts approx. 25 times as rapidly as Et2SO4, while at 0° the ratio is 58 to 1. The reaction in each case proceeds as far as the corresponding alkyl **hydrogen sulfate** or its Na salt, any further hydrolysis being negligible. These results demonstrate that the great difference in the velocities of reaction of alkali on the two alkyl sulfates is not mainly due to any difference of solubility on the part of the sulfates, because a similar difference is observed in homogeneous and heterogeneous system. The difference is, therefore, presumably to be attributed to the different character of the reactions, the alkali hydrolysis yielding the alkali **salt** of the alkylsulfuric and while the free alkylsulfuric acid is produced by the action of H2O or of alc. Examination of the reaction velocity of Et2SO4 and Me2SO4 with alc. in the presence of a gradually increasing portion of water shows that the former ester is distinctly less soluble in H2O and that the difference in the solubility of the 2 esters may exert an appreciable influence on the relative apparent activity of the two esters towards alkali hydroxide in the heterogeneous system. MeOH reacts with the 2 alkyl sulfates more rapidly than does EtOH, and although it was

found that, as expected, NaOMe affects the Me ester much more rapidly than the Et ester. the surprising result was obtained that MeONa in MeOH is less reactive than an EtOH solution of EtONa. An observation similar to this has already been made in certain cases. The suggestion of Kremann that the difference is due to the presence of traces of H₂O which causes a greater proportion of hydrolysis in the NaOEt is discredited and the suggestion is made that the explanation may be found in the possible occurrence of the reaction between the **alkyl sulfate** and the undissociated portion of the Na alkoxide.M

AB Me₂SO₄ undergoes hydrolysis by H₂O more rapidly than Et₂SO₄, but in the presence of KOH the ratio of the reaction velocities is very different from that observed for the hydrolysis by H₂O only. With 0.5 N KOH at 25°, the unimol. constant for MeOH is 45 times as great as for the Et₂SO₄, whereas with H₂O only the ratio is approx. 5:1. In order to decide whether the difference is due to the difference in the solubility of the 2 esters in H₂O, and to avoid the possibility of such a disturbing factor, it is desirable to exam. the rate of reaction in a homogeneous system. Kremann has already observed that with MeOH and EtOH, the rate of reaction of Me₂SO₄ is 3-4 times that of Et₂SO₄. With an alc. solution of EtONa, however, at 25° Me₂SO₄ reacts approx. 25 times as rapidly as Et₂SO₄, while at 0° the ratio is 58 to 1. The reaction in each case proceeds as far as the corresponding **alkyl hydrogen sulfate** or its Na salt, any further hydrolysis being negligible. These results demonstrate that the great difference in the velocities of reaction of alkali on the two alkyl sulfates is not mainly due to any difference of solubility on the part of the sulfates, because a similar difference is observed in homogeneous and heterogeneous system. The difference is, therefore, presumably to be attributed to the different character of the reactions, the alkali hydrolysis yielding the **alkali salt** of the alkylsulfuric and while the free alkylsulfuric acid is produced by the action of H₂O or of alc. Examination of the reaction velocity of Et₂SO₄ and Me₂SO₄ with alc. in the presence of a gradually increasing portion of water shows that the former ester is distinctly less soluble in H₂O and that the difference in the solubility of the 2 esters may exert an appreciable influence on the relative apparent activity of the two esters towards alkali hydroxide in the heterogeneous system. MeOH reacts with the 2 alkyl sulfates more rapidly than does EtOH, and although it was found that, as expected, NaOMe affects the Me ester much more rapidly than the Et ester. the surprising result was obtained that MeONa in MeOH is less reactive than an EtOH solution of EtONa. An observation similar to this has already been made in certain cases. The suggestion of Kremann that the difference is due to the presence of traces of H₂O which causes a greater proportion of hydrolysis in the NaOEt is discredited and the suggestion is made that the explanation may be found in the possible occurrence of the reaction between the **alkyl sulfate** and the undissociated portion of the Na alkoxide.M